



# Effect of group-3 elements doping on promotion of in-plane Seebeck coefficient of n-type Mg<sub>3</sub>Sb<sub>2</sub>

Chengliang Xia <sup>a</sup>, Juan Cui <sup>a</sup>, Yue Chen <sup>a, b, \*</sup>

<sup>a</sup> Department of Mechanical Engineering, The University of Hong Kong, Pokfulam Road, Hong Kong SAR, China

<sup>b</sup> HKU Zhejiang Institute of Research and Innovation, 1623 Dayuan Road, Lin An, 311305, China

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## ABSTRACT

Mg<sub>3</sub>Sb<sub>2</sub>-based alloys are promising thermoelectric materials with a reasonably low thermal conductivity. However, their electrical transport property is usually limited by the low carrier concentration. Mg<sub>3</sub>Sb<sub>2</sub> has a multi-valley conduction band with a six-fold degeneracy, benefiting n-type thermoelectric performance. Recently, n-type Y-doped Mg<sub>3</sub>Sb<sub>1.5</sub>Bi<sub>0.5</sub> and Sc-doped Mg<sub>3</sub>Sb<sub>2</sub>–Mg<sub>3</sub>Bi<sub>2</sub> alloys show a large figure of merit (ZT). In this paper, the doping effect of group-3 and chalcogen elements on the electronic structures and electrical transport properties of Mg<sub>3</sub>Sb<sub>2</sub> was investigated via the first-principles calculations. Chalcogen elements have a slight effect on the electronic structure, and Te-doped Mg<sub>3</sub>Sb<sub>2</sub> shows better normalized power factors in both the out-of-plane and in-plane directions, compared to the S-doped and Se-doped systems. Distinctly different doping effects appear in Mg<sub>3</sub>Sb<sub>2</sub> doped with group-3 elements. A increased density of states near the bottom of the conduction band can be induced by Sc or Y. Sc-doped and Y-doped Mg<sub>3</sub>Sb<sub>2</sub> show higher normalized power factors along the in-plane direction than those doped with chalcogens.

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## 1. Introduction

Thermoelectric materials have attracted much research interest because they can transform heat to electrical power or vice versa [1]. These materials are applied in narrow fields, such as green power generation and electronic industry [2,3]. Since most commercial thermoelectric materials, such as Bi<sub>2</sub>Te<sub>3</sub> or PbTe, are based on rare or toxic elements, some alternative materials need to be explored [4].

Bredt and Kendall firstly reported that Mg<sub>3</sub>Sb<sub>2</sub> is a potential thermoelectric material [5,6]. Mg<sub>3</sub>Sb<sub>2</sub> is a semiconductor with a small band gap of 0.8 eV [7], and it has intense alloying ability and structural stability [8,9]. Mg<sub>3</sub>Sb<sub>2</sub> has a reasonably low thermal conductivity, which is beneficial to the thermoelectric performance. It also has other advantages, such as, low mass density and low vapor pressure [6]. However, the poor electrical transport properties and the low carrier concentration (10<sup>17</sup> cm<sup>-3</sup>) of pristine

Mg<sub>3</sub>Sb<sub>2</sub> restrict its thermoelectric efficiency [10,11]. Condron et al. found that Mg<sub>3</sub>Sb<sub>2</sub> degrades at > 900 K and the maximum thermoelectric figure of merit (ZT) is 0.21 at 875 K [6].

Pristine Mg<sub>3</sub>Sb<sub>2</sub> is intrinsically p-type, and other p-type Mg<sub>3</sub>Sb<sub>2</sub>-based materials are also investigated extensively, such as Ag-doped and Na-doped Mg<sub>3</sub>Sb<sub>2</sub> [12–14]. Song et al. reported that the ZT of p-type Mg<sub>2.985</sub>Ag<sub>0.015</sub>Sb<sub>2</sub> reaches 0.51 at 725 K [13]. It is difficult to synthesize N-type Mg<sub>3</sub>Sb<sub>2</sub>-based materials due to the formation of Mg vacancies. In recent years, n-type Mg<sub>3</sub>Sb<sub>2</sub>-based materials with promising thermoelectric efficiencies are obtained by introducing Sb/Bi disorder and effective dopants, such as Mn and Te [7,11,15–17]. The figure of merit (ZT) of Te-doped Mg<sub>3</sub>Sb<sub>1.5</sub>Bi<sub>0.5</sub> at 300 K is 0.56, and it increases to 1.65 at 725 K. It was proposed that the improved thermoelectric performance of n-type Mg<sub>3</sub>Sb<sub>1.5</sub>Bi<sub>0.5</sub> is due to the multi-valley conduction band with a six-fold degeneracy and a light carrier effective mass.

Recent studies demonstrated that Y and Sc are effective n-type dopants in Mg<sub>3</sub>Sb<sub>2</sub> [18–20]. These dopants are found to increase the carrier concentration and enhance the carrier mobility of Mg<sub>3</sub>Sb<sub>2</sub> compound. Y-doped Mg<sub>3</sub>Sb<sub>1.5</sub>Bi<sub>0.5</sub> and Sc-doped Mg<sub>3</sub>Sb<sub>2</sub>–Mg<sub>3</sub>Bi<sub>2</sub> alloys both achieve a relatively high ZT value of 1.1 at 300–500 K. In addition, theoretically achievable carrier

\* Corresponding author. Department of Mechanical Engineering, The University of Hong Kong, Pokfulam Road, Hong Kong SAR, China.

E-mail address: [yuechen@hku.hk](mailto:yuechen@hku.hk) (Y. Chen).

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concentrations of group-3 elements in Mg<sub>3</sub>Sb<sub>2</sub> were also reported [21,22]. Nonetheless, the effects of Y and Sc on the electronic structure and electrical transport properties of Mg<sub>3</sub>Sb<sub>2</sub> are further investigated yet.

The thermoelectric energy conversion efficiency is described by the dimensionless figure of merit ZT, which is defined by

$$ZT = \frac{\sigma S^2}{\kappa} T \quad (1)$$

where  $\sigma$  is the electrical conductivity, S is the Seebeck coefficient, and  $\kappa$  is the thermal conductivity. High power factor and low thermal conductivity are required to achieve high-performance thermoelectric materials. Band engineering [23–26], e.g., the induction of resonant states and the increase of band convergence, is effective in enhancing the power factor and the thermoelectric figure of merit ZT. In this paper, we focus on a study on potential n-type dopants and their effects on the electronic structure and electrical transport properties since n-type Mg<sub>3</sub>Sb<sub>2</sub> has a higher thermoelectric efficiency,

## 2. Computational details

Density functional theory (DFT) [27,28] calculations were performed using the Vienna ab-initio simulation package (VASP) [29,30]. Effective band structures were computed by unfolding the electronic bands with BandUP [31]. Generalized gradient approximation (GGA) in the form of Perdew-Burke-Ernzerhof (PBE) parametrization was employed [32]. The TB-mBJ potential was also considered to correct the band gap [33]. The calculations of the electrical transport properties were performed based on the Boltzmann transport theory as implemented in BoltzTraP2 [34]. The constant relaxation time approximation was employed to investigate the electrical transport properties of doped Mg<sub>3</sub>Sb<sub>2</sub> systems.

We set an energy cutoff of 600 eV in the DFT calculations. Relaxation and charge self-consistent calculations were converged to 10<sup>-6</sup> eV. Cell and internal atomic coordinates both were fully relaxed until the force component was smaller than 10 meV/Å. The Brillouin zone was meshed with a density of  $2\pi \times 0.02 \text{ \AA}^{-1}$ , using  $\Gamma$ -centered Monkhorst-Pack grids. A denser  $k$ -point mesh of about  $2\pi \times 0.01 \text{ \AA}^{-1}$  was used to calculate the electrical transport properties. A  $2 \times 2 \times 2$  supercell containing 40 atoms, based on the Mg<sub>3</sub>Sb<sub>2</sub> P3m1 primitive cell, was constructed with one Mg or Sb atom substituted by one doping atom. For effective band structure calculations, atomic positions were relaxed after doping while the supercell vectors were fixed. For other calculations, supercells were fully optimized. The decomposed charge density of n-type Mg<sub>3</sub>Sb<sub>2</sub> was calculated by summing conduction electrons of all bands from the conduction band minimum to the energy level that corresponds to a carrier concentration of  $\sim 10^{20} \text{ cm}^{-3}$ . A larger supercell containing 90 atoms was used to verify the reliability of the results, as shown in Figs. S2 and S5 in the Supplemental Material.

Quantum Espresso [35] and EPW [36] were used to calculate the electronic relaxation time of Mg<sub>3</sub>Sb<sub>2</sub> based on DFT and density functional perturbation theory (DFPT). The electronic relaxation time was used in the calculation of the electrical transport properties of Mg<sub>3</sub>Sb<sub>2</sub> with BoltzTraP2 [34]. Norm-conserving relativistic pseudopotentials [37,38] were applied. A kinetic energy cutoff of 80 Ry and a  $14 \times 14 \times 8$   $k$ -mesh were used in the self-consistent calculations. The crystal structure was fully relaxed until the total energy was converged to 10<sup>-6</sup> Ry and the atomic forces were smaller than 10<sup>-5</sup> Ry/a.u. Phonons were calculated with a  $7 \times 7 \times 4$   $q$ -mesh. For the electron-phonon matrix element calculations, the  $14 \times 14 \times 8$   $k$ -mesh and  $7 \times 7 \times 4$   $q$ -mesh were interpolated to  $70 \times 70 \times 40$  fine  $k$ -

mesh and  $35 \times 35 \times 20$  fine  $q$ -mesh, respectively. The results calculated with different fine  $k$ -mesh and  $q$ -mesh are shown in Fig. S3 in Supplemental Material. A scissor shift was applied to correct the band gap to the experimental value of 0.8 eV [7].

## 3. Results and discussion

**Fig. 1a**) shows the trigonal crystal structure of Mg<sub>3</sub>Sb<sub>2</sub>, which has a space group of P3m1. Mg<sub>3</sub>Sb<sub>2</sub> is generally regarded as a Mg<sup>2+</sup> layer and a covalently bonded [Mg<sub>2</sub>Sb<sub>2</sub>]<sup>2-</sup> layer. There are three Wyckoff positions in Mg<sub>3</sub>Sb<sub>2</sub>, two Mg sites and one Sb site, and these positions are denoted in **Fig. 1a**). Mg1 and Mg2 represent Mg atoms in Mg<sup>2+</sup> layer and [Mg<sub>2</sub>Sb<sub>2</sub>]<sup>2-</sup> layer, respectively. The interstitial site (0, 0, 1/2) in Mg<sub>3</sub>Sb<sub>2</sub> lattice is also noted. **Fig. 1b**) and **Fig. 1c**) show the density of states and the band structure of pristine Mg<sub>3</sub>Sb<sub>2</sub> calculated with TB-mBJ potential. The theoretical band gap of Mg<sub>3</sub>Sb<sub>2</sub> is 0.65 eV, which is in reasonable agreement with the experimental band gap (0.8 eV) [7]. The valence band maximum locates at  $\Gamma$  point, and the conduction band minimum (CBM) locates in the path from M\*(0, 0.417, 0) to L\*(0, 0.417, 0.5). Zhang et al. demonstrated that the multi-valley band with a six-fold degeneracy along M\*-L\* and a duple valley conduction band at the K point promote the thermoelectric efficiency of n-type Mg<sub>3</sub>Sb<sub>2</sub> [11]. **Fig. 1d**) shows the electron-phonon scattering rate of Mg<sub>3</sub>Sb<sub>2</sub> near the band edges at 300 K and 600 K. The electronic relaxation time is calculated from the imaginary part of the electron self-energy. The scattering rate is the inverse of the electronic relaxation time [36].

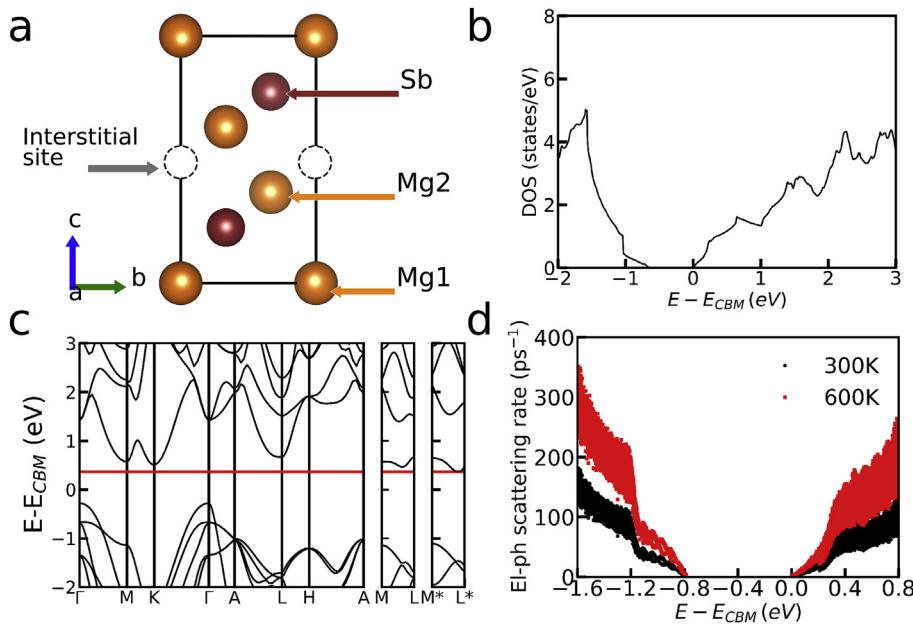
$$\frac{1}{\tau_{nk}} = 2\Sigma''_{nk}(\omega, T) \quad (2)$$

where  $\tau_{nk}$  is the electronic relaxation time of band  $n$  at wavevector  $k$ ;  $\Sigma''_{nk}(\omega, T)$  denotes the imaginary part of the electron self-energy. It is seen that the scattering rates of the electronic states close to the band edges are smaller, indicating weaker electron-phonon interactions. Electrons that are further away from the band gap can be more easily scattered to their final states [39].

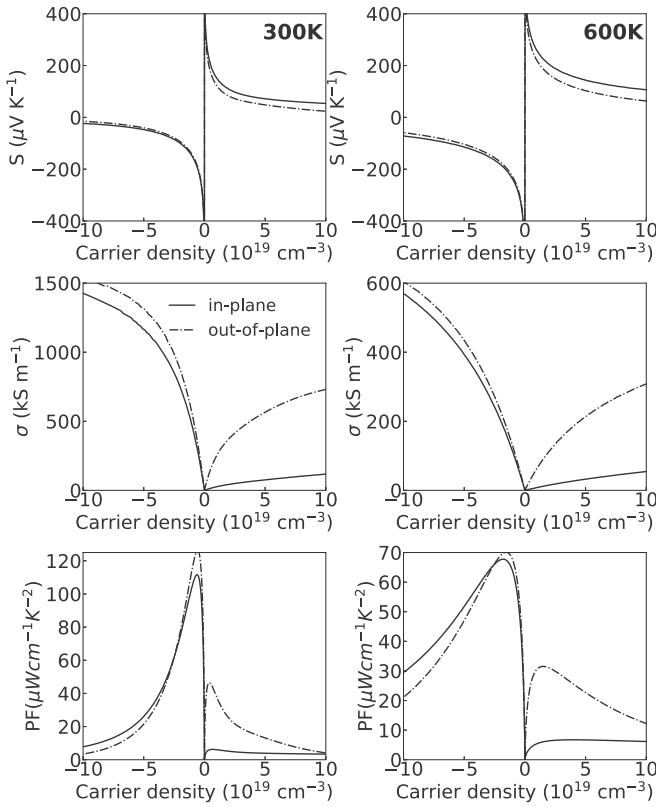
The Seebeck coefficient, the electrical conductivity, and the power factor of Mg<sub>3</sub>Sb<sub>2</sub> are calculated using EPW and BoltzTraP2, as shown in **Fig. 2**. Clearly, the power factor of n-type Mg<sub>3</sub>Sb<sub>2</sub> (negative carrier density) is generally larger than that of p-type Mg<sub>3</sub>Sb<sub>2</sub> (positive carrier density), indicating that n-type Mg<sub>3</sub>Sb<sub>2</sub> may exhibit a higher thermoelectric performance. Therefore, the calculations are consistent with previous experimental findings [11,16]. The calculated  $S$  and  $\sigma$  are, respectively, 2–3 times smaller and 4–6 times larger than the experimental data [13,40,41], being similar to previous electrical transport calculations of GeTe [39]. A difference between the theoretical and experimental results could be related to the defects in real materials.

Doping can be effective in improving the thermoelectric performance of Mg<sub>3</sub>Sb<sub>2</sub> [9,21,43]. In this work, the effect of group-3 (i.e., Sc and Y) and chalcogen group elements (i.e., S, Se and Te) doping are systematically investigated. It was found that group-3 elements (i.e., Sc and Y) tend to occupy the Mg1 lattice site, while the chalcogen group elements (i.e., S, Se and Te) tend to occupy the Sb lattice site [21,22]. Therefore, these doped Mg<sub>3</sub>Sb<sub>2</sub> systems are expected to be n-type because the doping atoms tend to donate one extra electron rather than the substituted atoms.

The density of states of doped and pristine Mg<sub>3</sub>Sb<sub>2</sub> was calculated to investigate the electronic structure changes. The conduction band minimum shifts to zero for direct comparison. Since the optimal carrier concentration of most thermoelectric materials is usually in the range of  $10^{19}$ – $10^{21} \text{ cm}^{-3}$ , the energy levels corresponding to a carrier density of  $\sim 10^{20} \text{ cm}^{-3}$  are shown in **Fig. 3a**). It is seen that the DOS of Sc-doped and Y-doped Mg<sub>3</sub>Sb<sub>2</sub> largely

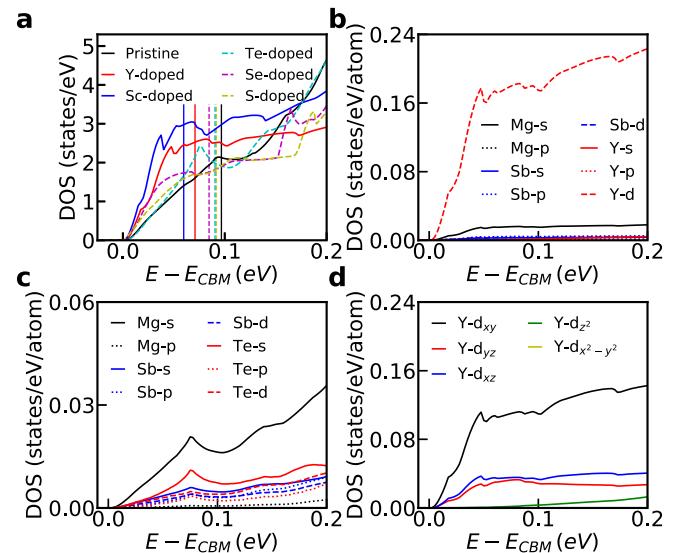


**Fig. 1.** **a)** Lattice structure of Mg<sub>3</sub>Sb<sub>2</sub>; the three Wyckoff positions and the interstitial site are denoted. **b)** Electronic density of states and **c)** band structure of pristine Mg<sub>3</sub>Sb<sub>2</sub> calculated with TB-mBJ potential. **d)** Electron-phonon scattering rate of Mg<sub>3</sub>Sb<sub>2</sub> with respect to the conduction band minimum at 300 K (black dots) and 600 K (red dots).



**Fig. 2.** The Seebeck coefficients (top), electrical conductivities (middle), and power factors (bottom) of Mg<sub>3</sub>Sb<sub>2</sub> at 300 K and 600 K. Hole and electron carrier concentrations are represented by positive and negative values, respectively.

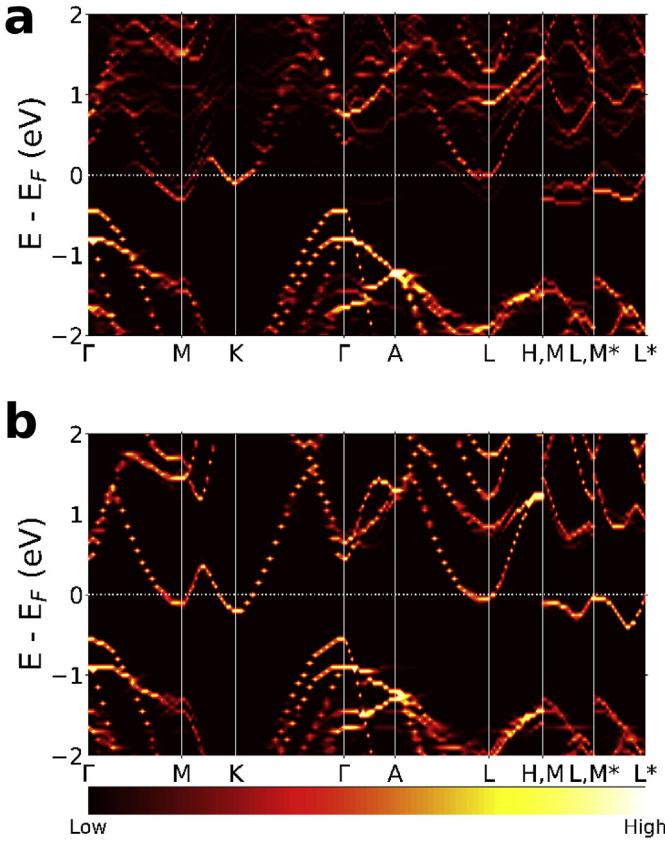
increases near the bottom of the conduction band. Also, for the systems doped with the chalcogen group elements (i.e., S, Se or Te), the change of DOS near the CBM is less significantly. To clarify the effects of group-3 and chalcogen group elements doping, we



**Fig. 3.** **a)** Density of states of R-doped (R = Y, Sc, S, Se or Te) and pristine Mg<sub>3</sub>Sb<sub>2</sub> supercells. The vertical lines represent the energy levels corresponding to a carrier concentration of  $-10^{20}$  cm<sup>-3</sup>. The conduction band minimum (CBM) shifts to zero. **b)** and **c)** Projected density of states per atom of Y-doped and Te-doped Mg<sub>3</sub>Sb<sub>2</sub>. **d)** The d-orbital projected density of states per atom of Y-doped Mg<sub>3</sub>Sb<sub>2</sub>.

calculated the projected density of states (PDOS) per atom of the representative Y-doped and Te-doped Mg<sub>3</sub>Sb<sub>2</sub> (see Fig. 3b) and Fig. 3c). Note that the DOS of Y-d states is dominantly larger than others, and a sharp increase of the DOS of Y-d states results in the overall increase of the total DOS. Furthermore, in Fig. 3d), the d<sub>xy</sub> orbital makes the major contribution to the Y-d PDOS near CBM. In contrast, after doping Te to the Sb lattice site, Mg-s states still dominate the DOS near CBM, and Te does not have a significant influence on the CBM.

Fig. 4 shows the unfolded effective band structures of Y-doped and Te-doped Mg<sub>3</sub>Sb<sub>2</sub>. Note that the band gaps of these effective



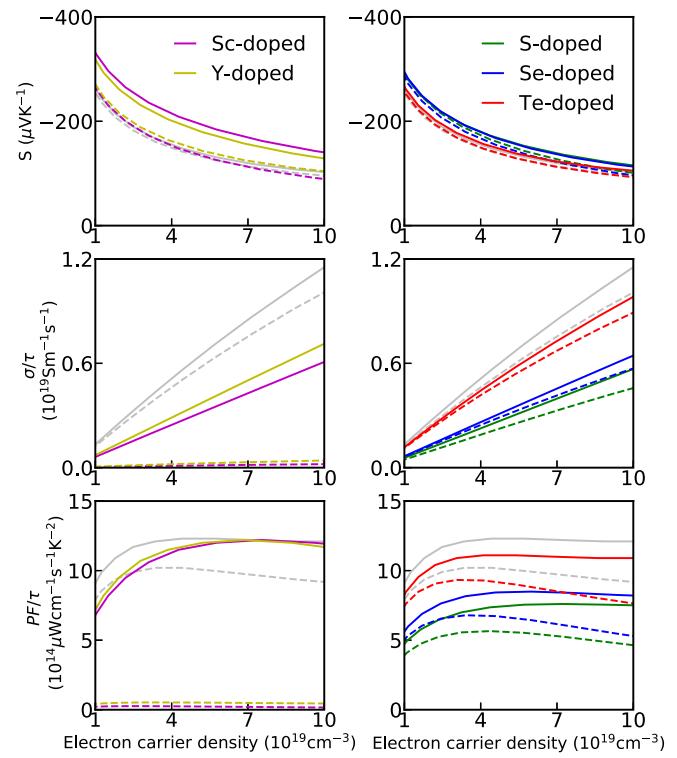
**Fig. 4.** Effective band structures of Y-doped (a) and Te-doped (b) Mg<sub>3</sub>Sb<sub>2</sub> supercells. The Fermi level is located at 0 eV.

band structures are underestimated due to the use of PBE functional. Compared to the band structure of pristine Mg<sub>3</sub>Sb<sub>2</sub>, relatively non-dispersive bands are formed in Y-doped Mg<sub>3</sub>Sb<sub>2</sub> along M-L near the bottom of the conduction band. This is believed to be related to the large increase of DOS near the CBM. In addition, the electronic band at M point moves down and becomes very close to the conduction band minimum, promoting the band convergence. By comparison, the effective band structure of Te-doped Mg<sub>3</sub>Sb<sub>2</sub> is similar to that of pristine Mg<sub>3</sub>Sb<sub>2</sub> except that the Fermi level moves into the conduction bands due to the extra electrons.

We calculated the electrical transport properties along both the in-plane and out-of-plane directions of doped Mg<sub>3</sub>Sb<sub>2</sub> supercells (see Fig. 5). The transport properties of pristine Mg<sub>3</sub>Sb<sub>2</sub> (40-atom supercell) are also calculated and shown by gray curves in Fig. 5 as a reference. The electrical transport properties calculated with a 90-atom supercell are consistent with those of the 40-atom supercell, as shown in Fig. S5 in Supplemental Materials. Also, Sc-doped and Y-doped Mg<sub>3</sub>Sb<sub>2</sub> are found to have larger Seebeck coefficients along the in-plane direction, which are consistent with their sharp increase of DOS near the CBM. From the analysis above, we found that the d<sub>xy</sub> orbitals of Sc and Y play an important role in enhancing the Seebeck coefficient along the in-plane direction. In contrast, chalcogen group elements (S, Se and Te) do not improve the Seebeck coefficient. The Seebeck coefficient can be described by the Mott relation [44]:

$$S = \frac{\pi^2 k_B^2 T}{3q} \cdot \frac{d[\ln(\sigma(E))]}{dE} \Big|_{E=E_F} = \frac{\pi^2 k_B^2 T}{3q} \cdot \left[ \frac{1}{n(E)} \frac{dn(E)}{dE} + \frac{1}{\mu(E)} \frac{d\mu(E)}{dE} \right] \Big|_{E=E_F} \quad (3)$$

where  $k_B$ ,  $\sigma(E)$ ,  $n(E)$  and  $\mu(E)$  are the Boltzmann constant, energy-



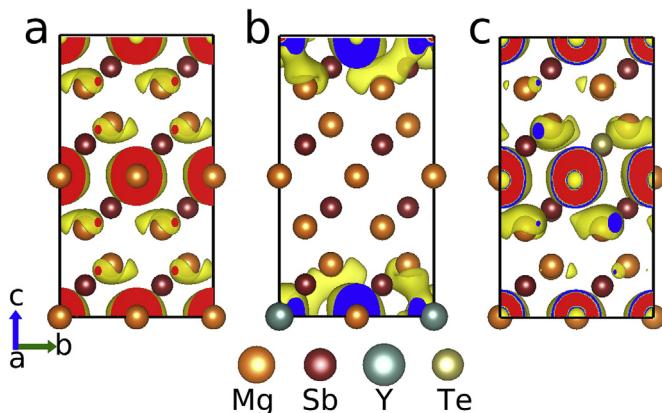
**Fig. 5.** The Seebeck coefficients (top), electrical conductivities normalized by the relaxation time (middle) and power factors normalized by the relaxation time (bottom) of doped and pristine Mg<sub>3</sub>Sb<sub>2</sub> supercells at 300 K. In-plane and out-of-plane directions are represented by solid and dashed curves, respectively. The transport properties of pristine Mg<sub>3</sub>Sb<sub>2</sub> (40-atom supercell) are denoted by gray curves as references.

dependent electrical conductivity, carrier concentration and mobility, respectively. Because Sc and Y induce a distortion in DOS near CBM and increase the DOS effective mass and  $\frac{dn(E)}{dE}$ , they promote the enhancement of the Seebeck coefficient.

When a Sc or Y atom occupies the Mg1 lattice site, the normalized electrical conductivity along the out-of-plane direction decreases dramatically, resulting in a very low normalized power factor along this direction. The normalized power factors of Sc-doped and Y-doped Mg<sub>3</sub>Sb<sub>2</sub> along the in-plane direction are greater than those of the chalcogen-doped systems. In the meantime, Te-doped Mg<sub>3</sub>Sb<sub>2</sub> has a higher normalized electrical conductivity and normalized power factor than the S-doped and Se-doped systems. To clarify the effect of dopants on the electronic structure, the decomposed charge densities corresponding to the bottom of the conduction band of pristine Y-doped, and Te-doped Mg<sub>3</sub>Sb<sub>2</sub> were calculated (see Fig. 6). The conduction electron distributions of Te-doped and pristine Mg<sub>3</sub>Sb<sub>2</sub> are rather similar. However, Y is found to change the electron distribution near the CBM dramatically. After doping Y, most conduction electrons along the out-of-plane direction disappear, which is consistent with the very low normalized electrical conductivity along this direction.

#### 4. Conclusions

We performed DFT calculations to investigate the effect of group-3 elements doping on the electronic structure and electrical transport properties of Mg<sub>3</sub>Sb<sub>2</sub>. Because n-type Mg<sub>3</sub>Sb<sub>2</sub> had a higher thermoelectric performance rather than the p-type in experiment, we focused on the group-3 and chalcogen group doping elements (i.e., Sc, Y, S, Se and Te). The group-3 dopants were



**Fig. 6.** Decomposed charge densities corresponding to the bottom of the conduction bands of pristine (a), Y-doped (b), and Te-doped (c)  $Mg_3Sb_2$  supercells for a carrier concentration of  $-10^{20} \text{ cm}^{-3}$ . The iso-surfaces correspond to a charge density of  $5 \times 10^{-5} \text{ e bohr}^{-3}$ .

found to have a much stronger effect on the CBM of  $Mg_3Sb_2$ , compared to the chalcogen group elements. The increase of DOS effective mass induced by doping group-3 elements resulted in a higher Seebeck coefficient along the in-plane direction. Also, the group-3 elements led to a large suppression of the out-of-plane normalized electrical conductivity. For the chalcogen group elements, Te-doped  $Mg_3Sb_2$  had a higher normalized power factor than the S-doped and Se-doped systems.

#### Declaration of competing interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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#### Appendix A. Supplementary data

Supplementary data to this article can be found online at <https://doi.org/10.1016/j.jmat.2020.03.003>.

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Yue Chen is an assistant professor at the University of Hong Kong, Department of Mechanical Engineering. His research interests focus on the materials physics for electrical and thermal transports, such as electronic structures and lattice dynamics. His interests stem from the studies of materials science at Oxford University and Beihang University. He was a postdoctoral fellow at Columbia University in the City of New York and Institute of Metal Research, Chinese Academy of Sciences before joining HKU.