

1 Effect of spacer configuration on the characteristics of
2 FO membranes: Alteration of permeation characteristics
3 by membrane deformation and concentration
4 polarization

5 Chulmin Lee^a, Jaewon Jang^a, Nguyen Thanh Tin^a, Suhun Kim^a, Chuyang Y. Tang^{b,c,d}, In S
6 Kim^{a,*}

7 ^a School of Earth Sciences and Environmental Engineering, Gwangju Institute of Science and
8 Technology (GIST), 123 Cheomdangwagi-ro, Buk-gu, Gwangju 61005, South Korea

9 ^b Department of Civil Engineering, the University of Hong Kong, Pokfulam, Hong Kong

10 ^c UNESCO Centre for Membrane Science and Technology, School of Chemical Engineering,
11 University of New South Wales, Sydney, New South Wales 2052, Australia

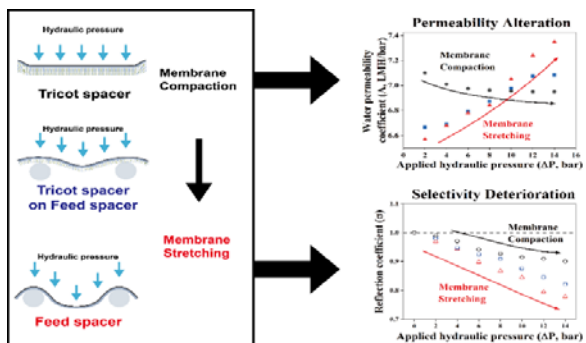
12 ^d UNSW Water Research Centre, School of Civil and Environmental Engineering, 12
13 University of New South Wales, Sydney, New South Wales 2052, Australia

14 *Corresponding author: School of Earth Sciences and Environmental Engineering, Gwangju
15 Institute of Science and Technology (GIST), 123 Cheomdangwagi-ro, Buk-gu, Gwangju 61005,
16 South Korea

17 +82-62-715-2436, iskim@gist.ac.kr

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21

22 **Abstract**

23 Membrane deformation is a significant problem in osmotically driven membrane processes, as
24 it restricts practical operating conditions and reduces overall process performance due to
25 unfavorable alteration of membrane permeation characteristics. In this respect, a spacer plays
26 a crucial role, as it dictates the form and extent of membrane deformation in association with
27 concentration polarization (CP), which is also influenced by spacer-induced hydrodynamic
28 behavior near the membrane surface. These two roles of spacers on membrane permeation
29 characteristics are inherently inseparable with the coexistence of hydraulic and osmotic
30 pressures. Here, we suggest a novel analytical method to differentially quantify the proportions
31 of effective osmotic pressure drop caused by membrane deformation and CP. Furthermore, we
32 tested two different FO membranes with three different spacer configurations to define and
33 discuss different forms of membrane deformation and their effects on membrane permeation
34 characteristics. The differential analysis revealed the effect of spacer configuration on effective
35 osmotic pressure drop in membrane deformation (up to ~ 201% of variation) is much greater
36 than that in CP (up to ~20.1% of variation). In addition, a combined configuration of a feed
37 spacer and tricot spacer demonstrated its ability of mitigating membrane deformation with
38 lower selectivity loss and channel pressure drop under pressurization.

39

40 **Introduction**

41 Osmotically driven membrane processes (ODMPs), such as forward osmosis (FO), pressure-
42 assisted forward osmosis (PAFO) and pressure-retarded osmosis (PRO), have been studied
43 extensively in recent years for water and energy-associated applications^{1,2}. The low-pressure
44 nature of FO and PAFO offers potential energy savings in niche applications where draw
45 solution regeneration is not required (or can be achieved using alternative energy sources with
46 low cost)^{1,3,4}. FO and PAFO also show more stable flux behavior than pressure-driven
47 membrane processes, as demonstrated by numerous bench-scale tests⁵⁻⁷. In parallel, bench-
48 scale PRO experiments often show highly attractive power densities for harvesting salinity
49 gradient power². Nevertheless, recent pilot tests⁸⁻¹² using commercial spiral wound modules
50 showed much poorer performance than expected due to deformation of the membranes and
51 spacer-filled channels inside the modules.

52 Detailed analysis revealed membrane deformation as one of the major critical limiting factors
53 during ODMP operation^{13,14}. In PRO, the pressure in the draw solution is often greater than 10
54 bar^{11,15}. Similarly, a maximum applied pressure of 6 bar has been reported for PAFO¹⁶. Even
55 in FO, where the applied pressure is theoretically zero, a pilot study reported a hydraulic
56 pressure of > 2 bar at the inlet to the draw channel for membrane elements arranged in series¹⁷
57 due to the hydraulic resistance of the flow channel. Unbalanced pressure across the membrane
58 can cause severe membrane deformation and blockage of flow channels (Figure 1a), which in
59 turn lead to greater pressure drop in the flow channels. This pilot study further demonstrated
60 that the maximum operable number of serially connected elements in FO operation is limited
61 by this draw channel pressure drop¹⁷. Furthermore, membrane deformation may also result in
62 severe deterioration of the separation properties as well as unfavorable mass transfer

63 conditions^{13,15,18}.

64 Concentration polarization (CP) is another inevitable phenomenon that significantly reduces
65 process performance in ODMPs by decreasing the effective osmotic pressure¹⁹. Membrane
66 deformation and CP are similar in that both are dictated by spacer geometry²⁰; however, they
67 are inherently dependent on each other and inseparable because hydraulic and osmotic
68 pressures always coexist regardless of process type in actual module-scale operation. To
69 critically understand the complex mechanisms of membrane deformation and CP, it is
70 necessary to quantitatively and differentially analyze effective osmotic pressure loss by these
71 two different causes because the extent, applied direction and ratio of osmotic and hydraulic
72 pressures are different for different operating conditions and processes.

73 In the current study, alteration of membrane separation properties and transport mechanisms is
74 investigated in association with differentiating membrane deformation and CP under different
75 spacer configurations. A conventional solution-diffusion (S-D) model-based characterization
76 method and thermodynamics-based novel characterization method are comparably analyzed to
77 reveal complex mechanisms of water and solute transport with regard to membrane
78 deformation and CP.

79

80

81 **Materials and Methods**

82

83 **Membranes**

84 Two commercial osmotic membranes were used in the current study. Thin-film composite
85 polyamide membrane PA-TFC (Toray Chemical Korea Inc., South Korea) consists of a
86 polyamide active layer, a polysulfone support layer, and a polyester support mesh embedded
87 in the polysulfone layer. Cellulose triacetate membrane CTA-ES2 was purchased from
88 Hydration Technology Innovations Inc. (US). This membrane also has an embedded polyester
89 support mesh, and one side of the membrane is thermally treated to form an active layer during
90 the fabrication process. Compared to PA-TFC, which has a relatively flat membrane surface
91 with a constant thickness of ~1.2 mm, CTA-ES2 has varying thickness in the range of
92 0.63~0.93 mm due to the embedded PET mesh (Figure S1).

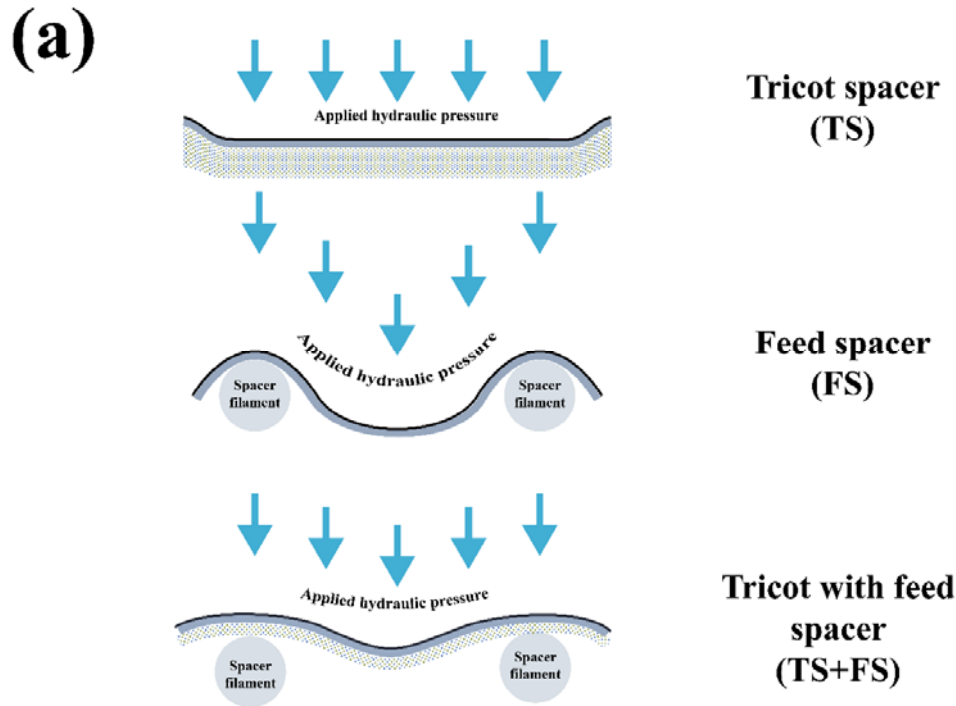
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94 **Spacer types and configurations**

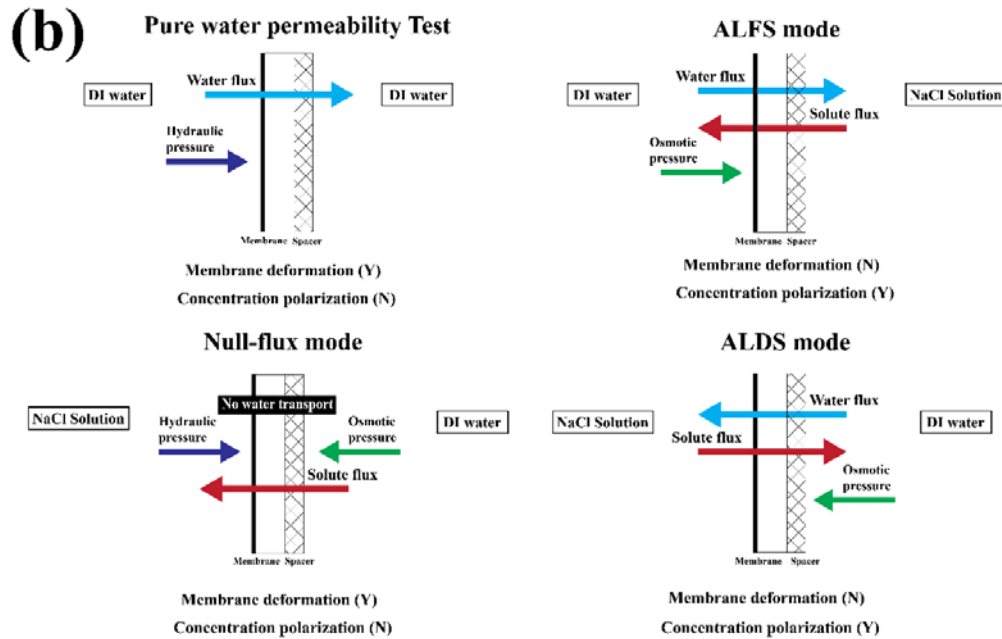
95 A diamond-shaped feed spacer (thickness of ~ 1.2 mm) and tricot fabric spacer (thickness of ~
96 0.2 mm) were both extracted from a spiral wound FO element (CSM FO-8040, Toray Chemical
97 Korea Inc., South Korea). The thickness of the feed spacer and tricot fabric spacer were
98 measured by using a digital caliper with a precision of 0.01 mm. The detailed dimensions and
99 images of these spacers are shown in Figures S1 and S3. Three spacer configurations were used
100 (Figure 1a): (i) tricot spacer only (TS), (ii) feed spacer only (FS), and (iii) both tricot spacer
101 and feed spacer (TS+FS). For a fair comparison, the total height of each spacer configuration
102 was kept constant as 1.4 mm for (FS: 1 feed spacer on top and 1 tricot spacer on the bottom,

103 TS+FS: 1 tricot spacer on top and 1 feed spacer on the bottom, TS: 7 sheets of tricot spacers).
104 The gap between spacer and channel height was filled by padding several plastic plates at the
105 bottom of the channel. These configurations lead to different spacer-induced membrane
106 deformations under pressurized conditions. Specifically, the membrane is more evenly
107 compressed onto the tricot spacer due to its flat geometry and smaller opening size (which
108 reduces the unsupported span of the membrane). In contrast, the membrane is stretched into
109 the relatively wide openings of the feed spacer. An intermediate deformation condition is
110 created by the combined use of the tricot spacer and feed spacer. According to previous
111 papers^{44,45}, this combined configuration was employed in a commercial 8 inch FO spiral-wound
112 element (Toray, CSM FO-8040). Throughout all the experiments a feed spacer was constantly
113 used to resolve the effect of spacer-induced axial channel pressure drop and CP in the feed
114 channel. The channel Reynolds number (Re_{ch})⁴⁶ computed based on feed spacer geometry in
115 this study was 33.42 (diameter of feed spacer filament is ~0.5 mm), which assures the flow
116 regime in this study falls into the laminar flow. Tricot spacer only configuration (TS) and
117 combined spacer configuration (TS+FS) are most likely to yield lower Re_{ch} due to its high
118 channel porosity.

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120



121

122 **Figure 1.** (a) Illustrations of spacer configurations used in this study and their associated
 123 membrane deformation; (b) schematic diagrams of the four operation modes and the presence
 124 of membrane deformation and CP in each mode.

125 Membrane performance tests

126 Figure 1(b) illustrates the four different operation modes used in this study (also see Supporting
127 Information S4 for a detailed schematic diagram of the laboratory-scale experimental setup):

- 128 • Pure water permeability test. In this test, deionized (DI) water was used in both feed
129 and draw solution channels to strictly eliminate the presence of osmotic pressure. To
130 take into account precompaction or prestretching of the membrane, each membrane was
131 pressurized under different spacer configurations for 2 h at each designated pressure
132 point. Then, the membrane flux was tested at 100%, 75%, 50% and 25% of the
133 designated pressure with a filtration duration of 5 min at each pressure step. For
134 example, when 2 bar was the designated pressure point, the membrane was first
135 pressurized under 2 bar for 2 h, and the water flux was measured immediately at 2.0,
136 1.5, 1.0 and 0.5 bar for 5 min, respectively. The water permeability was calculated as
137 the slope of these 4 points.
- 138 • FO tests in the active layer facing feed solution orientation (ALFS) and the active layer
139 facing draw solution orientation (ALDS). Both tests were performed under
140 nonpressurized conditions (less than 0.005 bar of channel pressure drop for both feed
141 and draw solution channels). NaCl solution and DI water were used as the draw solution
142 and feed solution, respectively. The water flux was driven by the osmotic pressure
143 difference across the membrane. In the ALFS, the draw solution is placed on the porous
144 substrate side such that a dilutive internal concentration polarization (ICP) occurs. In
145 contrast, the placement of the draw solution on the active layer side in the ALDS results
146 in a concentrative ICP that induces the reverse diffusion of the draw solutes into the

147 substrate. According to the literature, the ICP is more severe in the ALFS than the ALDS.
148 In the current study, water and solute fluxes were calculated from mass and conductivity
149 changes in the feed solution.

- 150 • Null-flux modes. In these modes, the membrane was characterized using null-flux mode,
151 where hydraulic pressure and osmotic pressure act in opposite directions to generate
152 zero water flux (i.e., zero change in water mass). For example, if 2 bar was the
153 designated pressure point, the equivalent draw solution concentration (0.0476 M) was
154 used to generate 2 bar of osmotic pressure. Then, to create zero water permeation,
155 hydraulic pressure was applied in the opposite direction of osmotically driven water
156 permeation. Since the water flux was zero, only the solute flux was measured in this
157 test mode. The Staverman reflection coefficient²¹ was derived using Equation S11 to
158 evaluate osmotic pressure loss solely by membrane deformation (no water transport
159 arguably induces negligible occurrence of CP). Before each test under null-flux mode,
160 2 h of membrane pressurization was conducted in designated pressure conditions to
161 consider the predeformation of the membrane.

162

163 Uniaxial tensile test for mechanical strength of membrane

164 To investigate the correlation between mechanical strength and spacer-induced membrane
165 deformation, a uniaxial tensile test was conducted using a universal testing machine (TEST
166 ONE, Product: TO-100-IC). Specimens of PA-TFC and CTA-ES2 membranes in dried and wet
167 states were prepared according to standard method³³ (ASTM D638: Standard Test Method for
168 Tensile Properties of Plastics) and tested at a displacement speed of 10 mm/min. Ultimate

169 tensile strength (UTS, MPa), elastic modulus (MPa) and elongation at break (%) were
170 computed based on the stress-strain curve of each specimen (Table S1).

171

172

173 **Results and Discussion**

174

175 **Alteration of permeation mechanism due to membrane deformation** 176 **and concentration polarization**

177 Pure water permeability is normally determined as the slope composed of multiple water flux
178 points measured at different hydraulic pressures using DI water. However, water permeability
179 can vary depending on the spacer and the extent of hydraulic pressure due to structural
180 deformation of the membrane and the resultant permeation mechanism. To examine
181 permeability variation depending on the hydraulic pressure and spacer configuration, a pure
182 water permeability test was conducted for the aforementioned three spacer configurations in
183 the given range of hydraulic pressure. Figure 2 illustrates an example of alteration of water
184 permeability (i.e., change in slope) under 2 hours of membrane compaction with hydraulic
185 pressures of 2 and 12 bar for PA-TFC and CTA-ES2 membranes. The TS configuration shows
186 a slight decline in the slope, while the slopes increase for the TS+FS and FS configurations.

187 For clear illustration, these water permeability changes are plotted in Figure 3(a) and 3(b).
188 Compared to the TS configuration, the TS+FS and FS configurations resulted in lower initial
189 permeability, which can be explained by the greater shadow effect (i.e., reduction in effective
190 membrane area) in the presence of spacers²⁹. For both PA-TFC and CTA-ES2 membranes, the
191 TS configuration shows a modest decreasing trend of water permeability. In contrast, the
192 TS+FS and FS configurations show a steep increasing trend. This disparity in performance
193 indicates that the TS and other two configurations have different forms of membrane
194 deformation. In the TS configuration, the membrane is evenly compacted onto the relatively

195 flat and rigid tricot spacer, leading to a densified membrane structure and thus increasing
196 membrane resistivity to water permeation. In the TS+FS and FS configurations, the membrane
197 elongates due to the open structure of the feed spacer. As a result of this tensile stretching, the
198 membrane becomes more permeable to water.

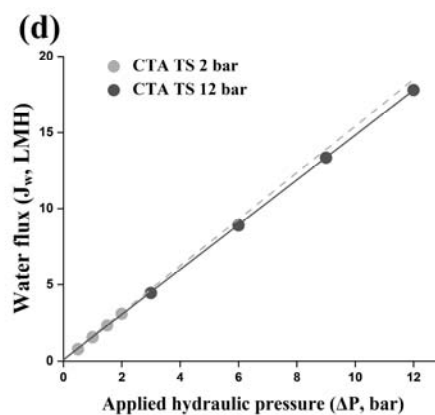
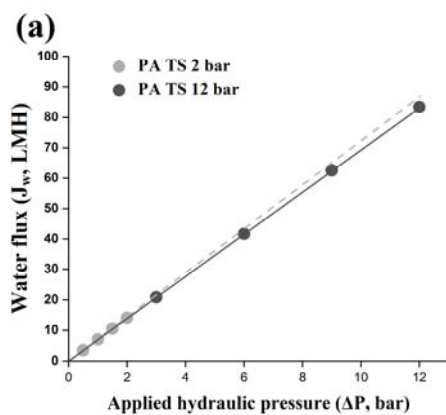
199 Although the two membranes show similar overall trends, their extents of variation were
200 noticeably different, particularly when the membranes were stretched. For example, the water
201 permeability of PA-TFC increased up to 12.9% in the FS configuration, which was much lower
202 than the corresponding change for CTA-ES2 (an increase of 55.5%) under identical testing
203 conditions. The better tolerance of PA-TFC to tensile stretching can be explained by the
204 flexibility of the polyamide active layer for tensile stretching³⁰.

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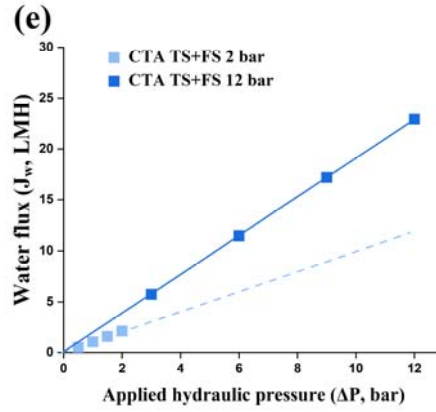
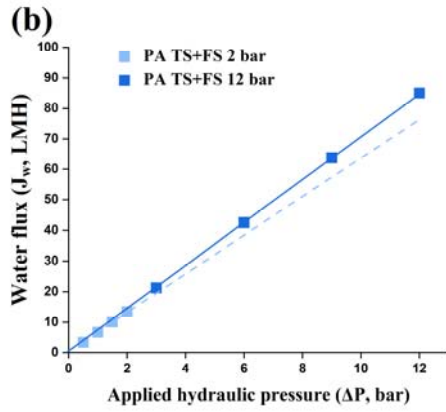
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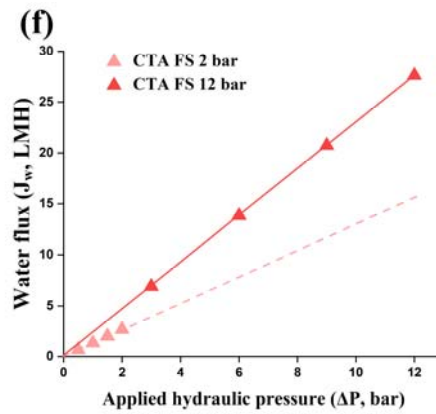
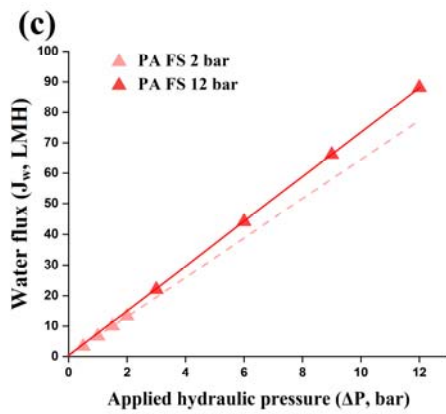
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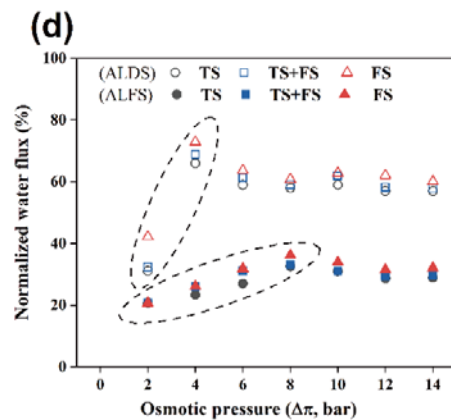
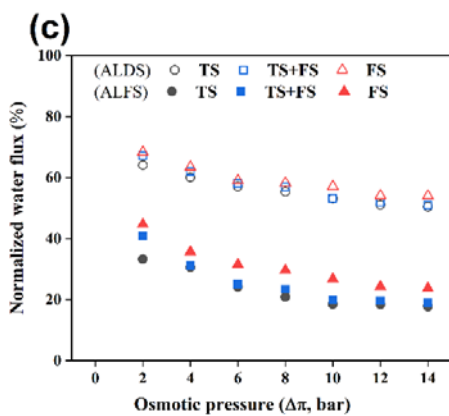
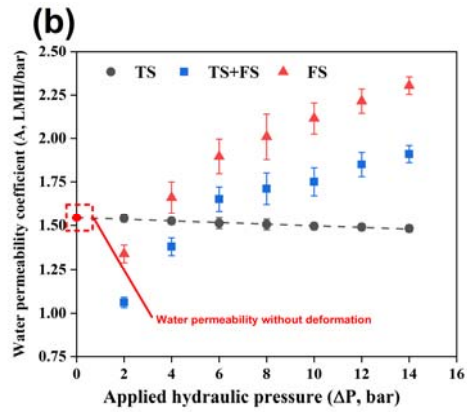
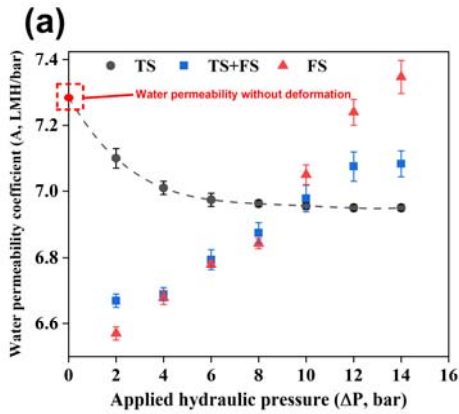
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212 **Figure 2.** Alteration of pure water permeability under effect of membrane compaction or
 213 stretching depending on spacer configuration for PA-TFC membrane (a, b and c) and CTA-ES2
 214 membrane (d, e and f). Pressure specified in legends represent hydraulic pressure employed
 215 for 2 hours of compaction/stretching depending on spacer configuration.

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219 **Figure 3.** Alteration of water permeability coefficient (A) under pressurized conditions for
 220 three spacer configurations, TS, TS+FS, and FS, for (a) PA-TFC and (b) CTA-ES2. Normalized
 221 water flux determined by pure water permeability in nondeformed conditions with different
 222 spacer configurations for (c) PA-TFC membrane and (d) CTA-ES2 membrane.

223

224 With current measurement technology, it is not viable to quantify the proportion of how much
 225 water flux declines due to CP solely by experimental methods. However, using an empirical
 226 relation between hydraulic pressure and water permeability, as shown in Figure 3(a), the water
 227 permeability under nondeformed conditions can be obtained. Using this water permeability in
 228 nondeformed conditions, the water flux in the ALFS and ALDS modes can be normalized to
 229 indicate how much water flux was reduced in each mode due to CP (Figure 2(c) and 2(d)). For

230 the PA-TFC membrane, the normalized water flux asymptotically decreases with increasing
231 bulk osmotic pressure. The TS and TS+FS configurations show relatively lower water flux than
232 that of the FS configuration, indicating that a dense tricot spacer (TS) located near the
233 membrane surface increases extent of CP, thus further reducing the effective osmotic pressure.
234 In addition, a feed spacer located in the bulk region does not reduce CP as effectively as the FS
235 configuration when a denser spacer is already located near the membrane surface (TS+FS).
236 This observation corresponds to other studies in which feed spacers in contact with membrane
237 surfaces are more likely to disturb the CP layer and thus enhance mass transfer^{20,25,26}. In the
238 ALDS mode, the water flux differences due to the spacer configuration were relatively small
239 compared to those in the ALFS mode, meaning that the importance of spacer selection in
240 reducing CP is more significant in dilutive internal concentration polarization (DICP) and
241 dilutive external concentration polarization (DECP) on the draw side than in concentrative
242 external concentration polarization (CECP) on the feed side.

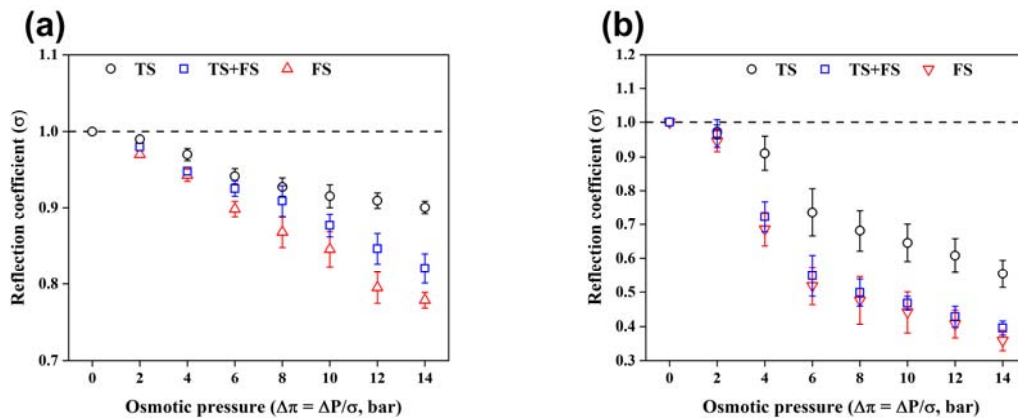
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244 Deterioration of selectivity of FO membrane due to structural 245 deformation

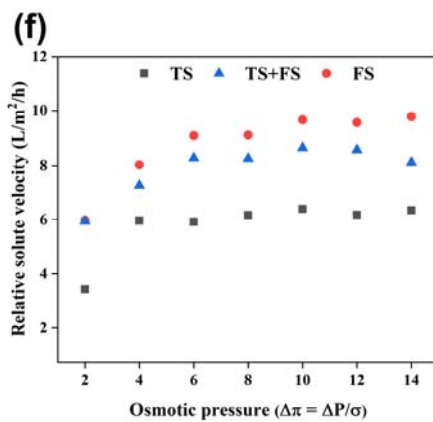
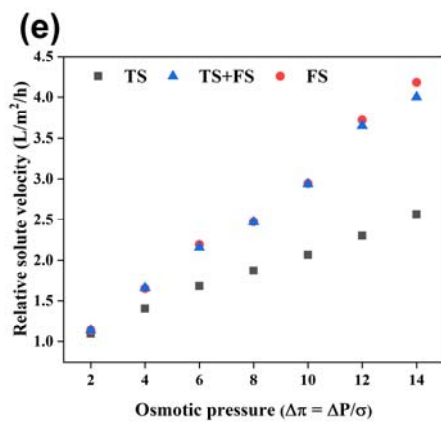
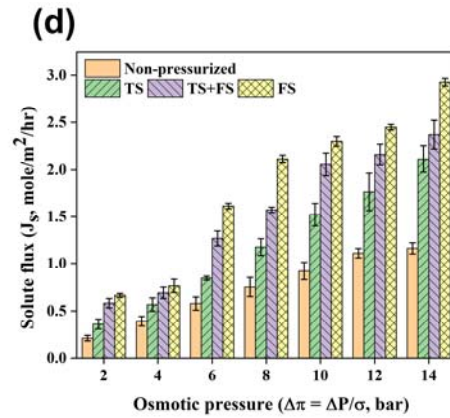
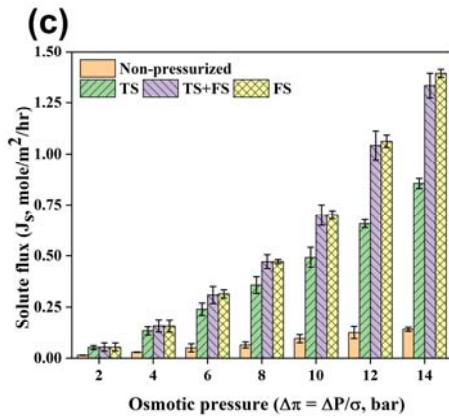
246 Generally, nonporous membranes such as FO and RO membranes are assumed to have a unity
247 reflection coefficient ($\sigma = 1$), which means that the membrane is completely selective to solute
248 species in nondeformed conditions. However, as hydraulic pressure engages the deformable
249 membrane structure, it loses its ability as a complete barrier, and accordingly, the effective
250 osmotic pressure decreases. This reduction in selectivity can be quantified by the reflection
251 coefficient (σ) as a function of corresponding osmotic pressure / adjusted hydraulic pressure
252 ($\Delta\pi = \Delta P/\sigma$) in each null-flux mode. Using Equation S11, the reflection coefficient was

253 derived in null-flux mode to examine the structural vulnerability of the membrane to hydraulic
 254 pressure. As shown in Figure 4(a) and 4(b), the CTA-ES2 membrane is highly vulnerable
 255 compared to the PA-TFC membrane in terms of a decreasing trend in the reflection coefficient.
 256 Membrane compaction (TS) has a much smaller effect on the reduction in the reflection
 257 coefficient than membrane stretching (TS+FS and FS) for both membranes. Here, it is
 258 intriguing to note that even under identical hydraulic pressure and spacer configurations, each
 259 membrane corresponds differently in terms of the reflection coefficient. For example,
 260 compared to the FS configuration, the TS+FS configuration exhibits some partial effectiveness
 261 in reducing the reduction in the reflection coefficient for PA-TFC (Figure 4(a)) but shows little
 262 effect for CTA-ES2. This result suggests that using the TS+FS configuration may not be
 263 sufficient to avoid membrane deformation, particularly for the CTA-ES2 membrane.

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268 **Figure 4.** Reduction in the reflection coefficient for three spacer configurations, TS, TS+FS,
 269 and FS, for (a) PA-TFC and (b) CTA-ES2. Comparison of solute flux under the nonpressurized
 270 ALDS mode and null-flux mode for (c) PA-TFC and (d) CTA-ES2. Relative solute velocity
 271 derived by normalizing the solute flux by the respective NaCl concentration for (e) PA-TFC
 272 and (f) CTA-ES2. Figure 4(c) and 4(d) compare the solute fluxes under the nonpressurized
 273 ALDS mode and null-flux mode for the three spacer configurations.

274

275 The CTA-ES2 membrane shows higher solute flux in all operation modes than the PA-TFC
 276 membrane. However, the ratio of ~~in~~ the solute flux under pressurized mode (null-flux mode)
 277 over that under nonpressurized mode and pressurized mode (null-flux mode) was significantly
 278 higher for PA-TFC (approx. 10-fold) than for CTA-ES2 (approx. 2-fold). This increased solute
 279 transport under pressurized mode can be explained by two factors: (i) convective solute

280 transport through defects in the active layer formed by membrane deformation and (ii)
281 enhanced solute diffusion by the higher effective concentration difference under null-flux mode
282 (due to negligible CP under this condition). Based on the observation of different solute fluxes
283 under membrane compaction (TS) and membrane stretching (TS+FS and FS), it is evident that
284 membranes become more permeable to solute species with tensile stretching.

285 The solute flux increases as a function of the concentration difference across the membrane
286 where there is no water transport ($J_w = 0$)³¹. To remove the concentration dependence, the
287 relative solute velocity was derived by normalizing the solute flux by the corresponding NaCl
288 concentration based on Equation S13 (Figure 4(e) and (f)). The slopes of the plots in these
289 figures show that increasing solute transport is solely based on selectivity deterioration as a
290 function of hydraulic pressure, excluding the effects of CP and NaCl concentration. Thus, this
291 slope can be used as a coefficient of membrane selectivity deterioration due to spacer-induced
292 membrane deformation. The relative solute velocity of the CTA-ES2 membrane showed a
293 significant initial increase up to 6 bar but then leveled off without a significant increase, while
294 that of PA-TFC showed a constant increasing trend. This most likely means that the active layer
295 of CTA-ES2 was already fully stretched (i.e., losing integrity for selectivity) at 6 bar of
296 hydraulic pressure regardless of the spacer configuration, but the active layer of PA-TFC was
297 relatively more flexible to show a constant increasing trend in the given pressure range.
298 Differences between the TS+FS and FS configurations were found above 4 bar for CTA-ES2
299 and 12 bar for PA-TFC. It should be noted that loose membranes such as CTA-ES2 may not be
300 appropriate for this analysis since they exhibit severe nonlinearity, as shown in Figure 3(f).

301

302 Relation between mechanical strength of membrane and structural
303 vulnerability

304 Mechanical strength of membrane is directly related to the susceptibility of the membrane
305 against spacer-induced membrane deformation, and resulted membrane's permeation
306 characteristics. Several studies²²⁻²⁴ examined the mechanical strength of PA-TFC and CTA-ES
307 membranes in a dried state, however, the mechanical strength of the membranes in wet-state
308 has not been investigated. Although it is common to use dried specimens for mechanical
309 analysis, the mechanical strength of the wet membrane also needs to be analyzed since FO
310 membrane's permeation characteristics are measured in a wet state. To comparatively examine
311 the mechanical strength of PA-TFC and CTA-ES2 membranes in a dried and wet state, the
312 uniaxial tensile test was conducted. As shown in Table. S1, ultimate tensile strength (UTS) of
313 the PA-TFC membrane was somewhat increased to 66.2 ± 3.18 MPa after wetting in water
314 compared to the dried state (66.6 ± 3.94 MPa), whereas UTS value of the CTA-ES2 membrane
315 was 47.6 ± 0.8 MPa in a dried state, and it was decreased to 44.1 ± 1.2 MPa after wetting in
316 water. The higher UTS value of the wet PA-TFC membrane than that of the dried PA-TFC
317 membrane is not the behavior of typical polymeric materials. This result can be originated from
318 the unique structure of the PA-TFC membrane. Because of the polyester mesh is embedded
319 inside the polysulfone, water molecules can interact as a cross-linker by hydrogen bonding
320 between the polymers when the membrane is wet^{28,33-36}. The increased elastic modulus of the
321 wet PA-TFC membrane can also be considered in connection with the phenomenon described
322 above. The elastic modulus of dried and wet PA-TFC membrane was 716.7 ± 57.26 and
323 736.3 ± 74.87 MPa, respectively. And, the behavior of the membranes was almost same in
324 before and after wetting, which means that water molecules affect the strength of interactions

325 between materials and structural stability in the region of elastic deformation without changing
326 the intrinsic physical properties of the polymers. The elastic modulus of the dried and wet CTA-
327 ES2 membrane was 824.8 ± 167.47 and 531.8 ± 102.25 MPa, respectively. This is a common
328 result that the mechanical strength is weakened because the structure loosens when the polymer
329 or organic material is wet with water^{37,38}. This drastic reduction of mechanical strength and
330 inherent stiffness of CTA-ES2 membrane in wet state explains high susceptibility of CTA-ES2
331 membrane to hydraulic pressure (i.e. severe reduction of reflection coefficient under
332 pressurization). The evidence of the structural vulnerability of CTA-ES2 membrane can also
333 be found in visual observation. Membrane surfaces before-and-after pressurization analyzed
334 by scanning electron microscopy (SEM) (Figure S2 and S3 for PA-TFC and CTA-ES2
335 membrane) reveals that CTA-ES2 membrane showed severer structural damages than PA-TFC
336 membrane on both active and support layers more. Furthermore, more visible damages with
337 membrane stretching (FS) than with membrane compaction (TS) for both membranes.
338 Although, PA-TFC membrane turned out to be more vulnerable specifically for membrane
339 stretching by feed spacer on the active layer as it shows evident rupture by feed spacer filament
340 (Figure S2(e)) This can be attributed to structural differences in the active layer and
341 intermediate layer between the active and support layers. Unlike a PA membrane, which is
342 composed of a polysulfone support layer and a thin polyamide active layer bonded by
343 interfacial polymerization (IP), a CTA membrane is composed of a single component, and the
344 active layer is formed by thermal treatment of the support layer. Due to this relatively weak IP-
345 bonded interface compared to the thermally treated interface in a CTA membrane²⁷, a PA
346 membrane can be more vulnerable to structural deformation of the active layer. In addition, the
347 flexibility of the polyamide active layer³⁰ allows the solute flux of the PA-TFC membrane to
348 be more varied than that of the CTA-ES2 membrane. In this respect, the PA-TFC membrane

349 shows a higher solute flux increment ratio (approx. 10 times) between nonpressurized and
350 pressurized conditions than the CTA-ES2 membrane (approx. 2 times), as shown in Figure 4(c)
351 and (d).

352 The other important aspect of membrane deformation is channel pressure drop due to the
353 contracted channel geometry under pressurized conditions. At present, no method has been
354 developed to quantify this channel contraction and resulted reduction in channel volume.
355 However, this channel contraction can be indirectly estimated by the channel pressure drop
356 measured in deformed channel conditions. As shown in Figure S5, CTA-ES2 membrane
357 revealed a higher channel pressure drop than that of PA-TFC membrane, which can be
358 attributed by the aforementioned drastic reduction of mechanical strength and stiffness of CTA-
359 ES2 membrane in a wet state.

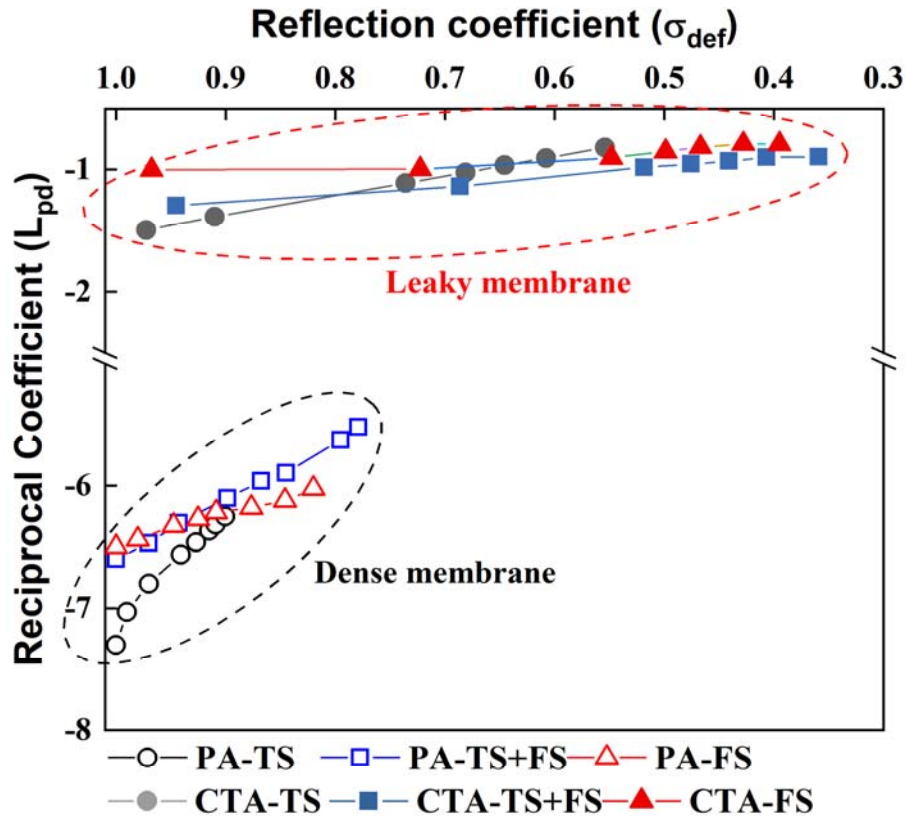
360

361 Change in interdependence between water and solute transport

362 To assess the interdependence between volumetric flux (J_v) and relative solute velocity (J_s),
363 the experimentally obtained reflection coefficient was plotted as a function of the reciprocal
364 coefficient that is derived by reciprocal relations (Equations S6 and S7). Figure 5 depicts this
365 interdependence for the PA-TFC and CTA-ES2 membranes for different spacer configurations.
366 Throughout the given range of the reflection coefficient, membrane compaction (TS) induces
367 a lower reciprocal coefficient than membrane stretching (TS+FS and FS) because membrane
368 stretching makes the membrane more vulnerable to solute transport, as shown in previous
369 results. The FS configuration seems to have a lower reciprocal coefficient than the TS+FS
370 configuration in the low-pressure range but exceeds the TS+FS configuration above 6 bar with

371 a steep increase in this coefficient. This finding is most likely due to the initial difference in
372 water permeability caused by the shadow effect, and the shadow effect is then dominated by
373 the effect of membrane stretching above 6 bar. On the other hand, CTA-ES2 reveals a
374 drastically lower reciprocal coefficient than PA-TFC with a relatively small deviation. As
375 depicted in Figure 5, the reflection coefficient of CTA-ES2 decreases much more rapidly than
376 that of PA-TFC, indicating the different pressure susceptibilities of each membrane. Since a
377 higher negative value of the reciprocal coefficient means more ideal membrane conditions (less
378 interaction between water and solute transport), the CTA-ES2 membrane can be defined as a
379 highly leaky membrane with high pressure vulnerability compared to PA-TFC, which can be
380 defined as a dense membrane.

381



382

383 **Figure 5.** Reciprocal coefficient as a function of the reflection coefficient for PA-TFC and
 384 CTA-ES2.

385

386 Characteristics of FO membranes: S-D model and irreversible
 387 thermodynamic model

388 There are several blind spots in general S-D model-based characterization methods as follows:

- 389 (i) A single permeability coefficient is assumed regardless of spacer and hydraulic pressure
 390 conditions; (ii) No interaction between solute and water transport is considered; (iii) A
 391 nonporous membrane and only diffusion transport of water and solute are assumed. In the
 392 current study, alteration of water and solute permeability was demonstrated under both

393 pressurized and nonpressurized conditions for three spacer configurations. As demonstrated in
394 the results, alteration of water and solute permeability is significant and varies depending on
395 the spacer configuration in a given hydraulic pressure range. These deviations may be
396 negligible in an RO process since these discrepancies would be insignificant under high
397 operating pressures (40~60 bar) with a single-spacer configuration (i.e., tricot spacer). However,
398 when considering that ODMPs use relatively various spacer configurations under low
399 operating pressures (below 20 bar), this deviation of the permeability coefficients needs to be
400 considered in precisely estimating and analyzing process performance. The interdependence
401 between solute and water transport was quantified by the reciprocal coefficient as a function of
402 the reflection coefficient shown in Figure 5. This unique characterization method based on
403 thermodynamic assessment can provide a new aspect regarding transport mechanisms in
404 membrane processes. However, the literature covering this aspect is severely limited at the
405 moment in terms of discussion, and further experimental validation is required for the validity
406 of this approach.

407 On the other hand, a drawback of thermodynamic assessment is that it does not provide insights
408 regarding the physical and chemical status of the solution inside and near the membrane, while
409 an S-D model-based method attempts to draw a concentration profile in association with the
410 structural characteristics of the membrane and flow characteristics in the flow channel (i.e.,
411 ICP and ECP equations). However, it should not be forgotten that all these theoretical works
412 can only be validated by the resultant volume and concentration changes in solutions during
413 process operation. Furthermore, the structural parameter (S) and mass transfer coefficient (k),
414 which are the most important factors for addressing ICP and ECP, respectively, cannot be
415 experimentally validated. For example, as shown in Equation S5, the structural parameter is

416 theoretically dependent on the thickness, tortuosity and porosity of the support layer, but in
417 most studies, this S value is derived from predetermined A and B values using Equation S3 due
418 to difficulties in nanoscale measurements of those structural characteristics. The mass transfer
419 coefficient can be derived using the relation given in Equations S6 and S7; however, this
420 approach may not be suitable to precisely resolve a complicated geometry of spacer-filled
421 narrow channel and the resulted complex microflow near the membrane surface.

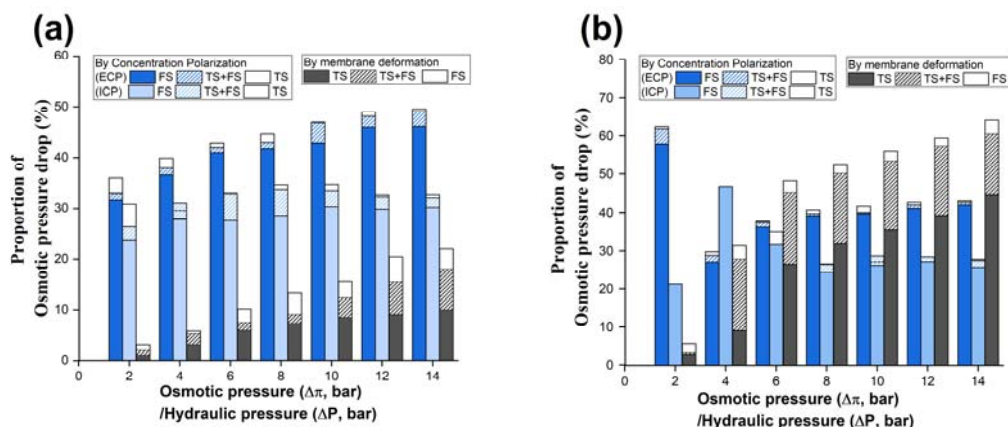
422

423 Effect of spacer configuration on membrane deformation and CP

424 As discussed in the previous section, it is not viable to draw a concentration profile across the
425 membrane layers solely based on experimentally measured values. However, the cause-specific
426 proportions of the osmotic pressure drop can be derived purely based on phenomenological
427 assessments using three assumptions: (i) In the nonpressurized ALDS mode, the effect of ICP
428 can be neglected (i.e., the core assumption in nonpressurized characterization methods for FO
429 membranes³²); (ii) The nondeformed water permeability derived by empirical relation (Figure
430 3) can be postulated as water permeability excluding the effects of membrane deformation and
431 CP; (iii) The Staverman reflection coefficient measured under null-flux mode represents the
432 osmotic pressure drop solely by membrane deformation. Although the effects of deformation
433 and CP cannot be directly compared due to differences in their mechanisms, the proportions of
434 the osmotic pressure drop by CP and deformation can be compared as a function of the same
435 extent of osmotic pressure and hydraulic pressure differences. Figure 6(a) and (b) compares
436 the proportion of osmotic pressure drop caused by CP and membrane deformation depending
437 on spacer configuration. The increment ratio of CP-induced osmotic pressure drop by different
438 spacer configurations varied from 7.98% to 20.7% for PA-TFC membrane and from 0.19% to
439 4.76% for CTA-ES2 membrane. Theoretically, CP is the resultant of the membrane's

440 equilibrium in association with water and solute transports, yet mostly due to water transport³⁹.
 441 In this respect, the difference in the increment ratio is attributed to the different water
 442 permeability of the two membranes (~5-fold difference). On the other hand, the deformation-
 443 induced proportion of osmotic pressure drop by spacer configuration increased from 72.3% to
 444 200% and 18.2% ~ 201% for PA-TFC and CTA-ES2 membrane respectively. This difference
 445 in variation of osmotic pressure drop is due to different forms and extent of membrane
 446 deformation and its susceptibility of each membrane as discussed in previous sections. Overall,
 447 the above comparison of osmotic pressure drop caused by CP and membrane deformation
 448 reveals the influence of spacer is much greater in membrane deformation than that in CP.

449



450

451 **Figure 6.** Comparison of proportions of osmotic pressure drop by CP and membrane
 452 deformation (PA-TFC for (a) and CTA-ES2 for (b)).

453

454

455 **Implications**

456 Membrane deformation and CP are the most dominant factors that reduce effective osmotic
457 pressure in ODMPs. Although these two factors are mutually dependent when hydraulic
458 pressure and osmotic pressure coexist, which is an inevitable circumstance in module-scale
459 operation, the two factors were differentiated and comparatively quantified using extensive
460 theoretical and experimental analyses for a deeper understanding of the permeation
461 characteristics of FO membranes. It was demonstrated that membrane deformation, especially
462 membrane stretching and the resulting selectivity loss, can be controlled by a combined
463 configuration with a feed spacer and tricot spacer. The results also revealed that the role of a
464 spacer in reducing deformation-induced osmotic pressure drop is more significant than that in
465 reducing CP-induced osmotic pressure drop under identical osmotic and hydraulic pressures.
466 Although most spacer research has tended to focus on enhancing mass transfer by reducing CP,
467 the results in this study imply that mitigation of membrane deformation and the resulted
468 deterioration of process performance need to be considered more in membrane and spacer
469 studies. The membrane characterization method suggested in this study can be utilized to
470 quantitatively and differentially evaluate the effects of membrane deformation and CP on
471 effective osmotic pressure drop for the development of ODMP membranes and spacers.

472 Although the current study focuses primarily on the effect of spacer configuration on
473 membrane deformation and CP, spacer also have major effect on fouling. For example, spacer
474 geometry can directly affect the fluid dynamics of spacer-filled channels to mitigate foulant
475 deposition⁴⁰. However, in the context of ODMPs, severe spacer-induced deformation may
476 block the flow channels and result in severe membrane fouling. In addition, the increased
477 reverse solute diffusion (e.g., of fouling precursors such as Ca^{2+})⁴¹ as a result of severe

478 membrane deformation can accelerate fouling and scaling^{42,43}. Future studies need to
479 systematically investigate these competing effects of spacers on fouling of ODMP membranes.

480

481 **Supporting Information**

482 SEM images of cross-section and surface of PA-TFC and CTA-ES2 membrane before and after
483 pressurization, images of feed spacer and tricot spacer, mechanical properties of PA-TFC and
484 CTA-ES2 membrane in dried and wet state, detailed experimental set-up for pressurized and
485 nonpressurized tests, experimental results on channel pressure drop, and theoretical
486 backgrounds regarding FO membrane characterization with the effect of the spacer.

487

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492

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1 Supporting information

2
3 Effect of spacer configuration on characteristics of FO
4 membrane: Alteration of permeation characteristics by
5 membrane deformation and concentration polarization

6 Chulmin Lee^a, Jaewon Jang^a, Nguyen Thanh Tin^a, Suhun Kim^a, Chuyang Y. Tang^{b,c,d}, In S Kim^a

7 a School of Earth Sciences and Environmental Engineering, Gwangju Institute of Science and
8 Technology (GIST), 123 Cheomdangwagi-ro, Buk-gu, Gwangju 61005, South Korea

9 b Department of Civil Engineering, the University of Hong Kong, Pokfulam, Hong Kong

10 c UNESCO Centre for Membrane Science and Technology, School of Chemical Engineering,
11 University of New South Wales, Sydney, New South Wales 2052, 10 Australia 11

12 d UNSW Water Research Centre, School of Civil and Environmental Engineering, 12
13 University of New South Wales, Sydney, New South Wales 2052, Australia

14 Corresponding author: +82-62-715-2436, iskim@gist.ac.kr

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45 Membrane and spacer properties

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54 Figure S1. SEM images of cross-section of pristine PA-TFC (a) and CTA-ES2 (b) membrane

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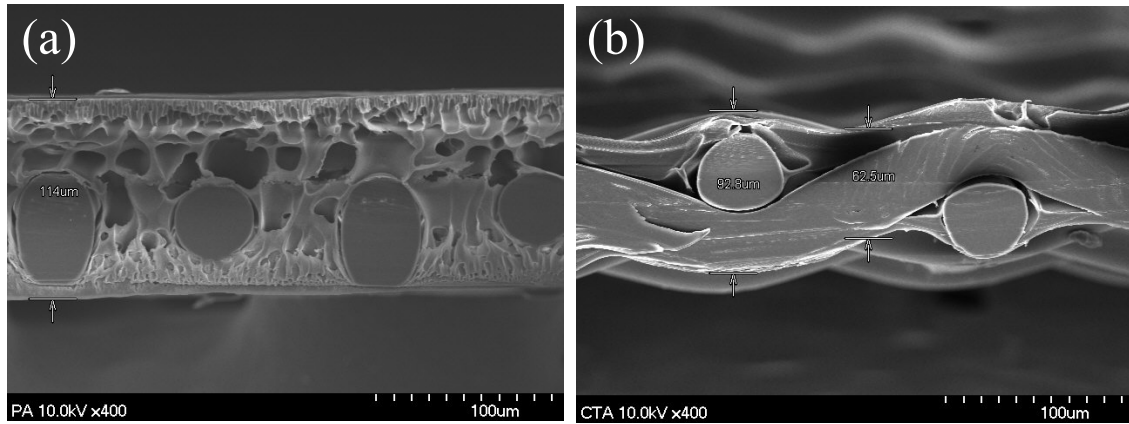
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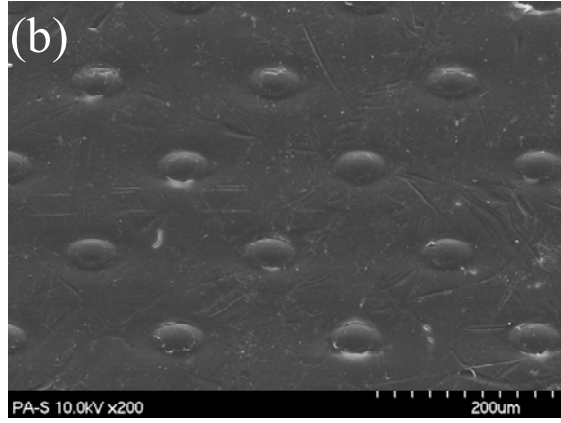
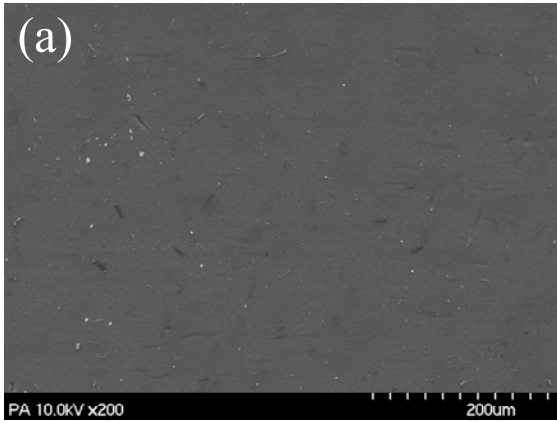
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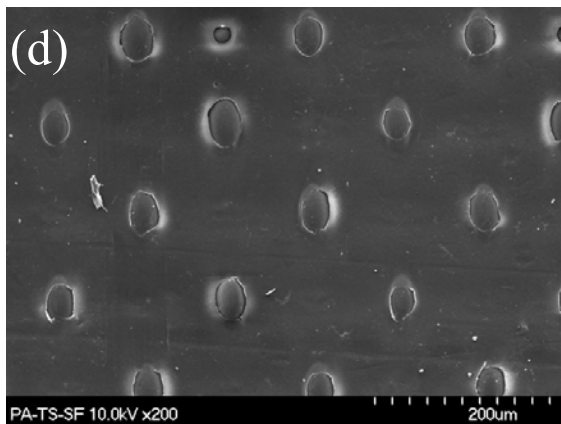
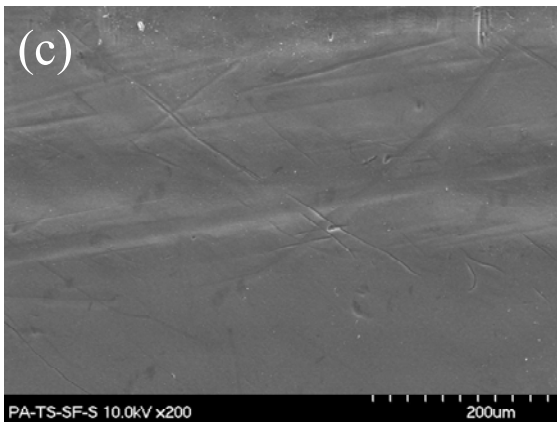
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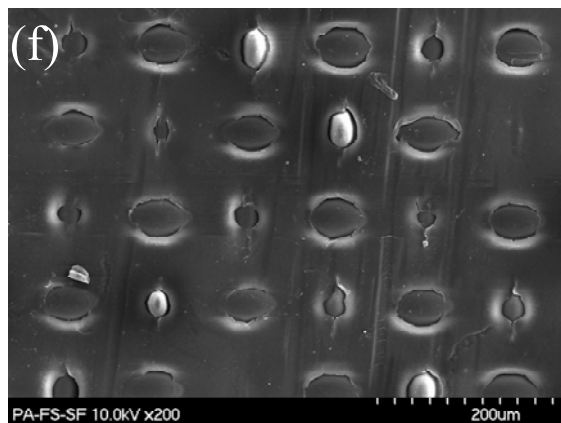
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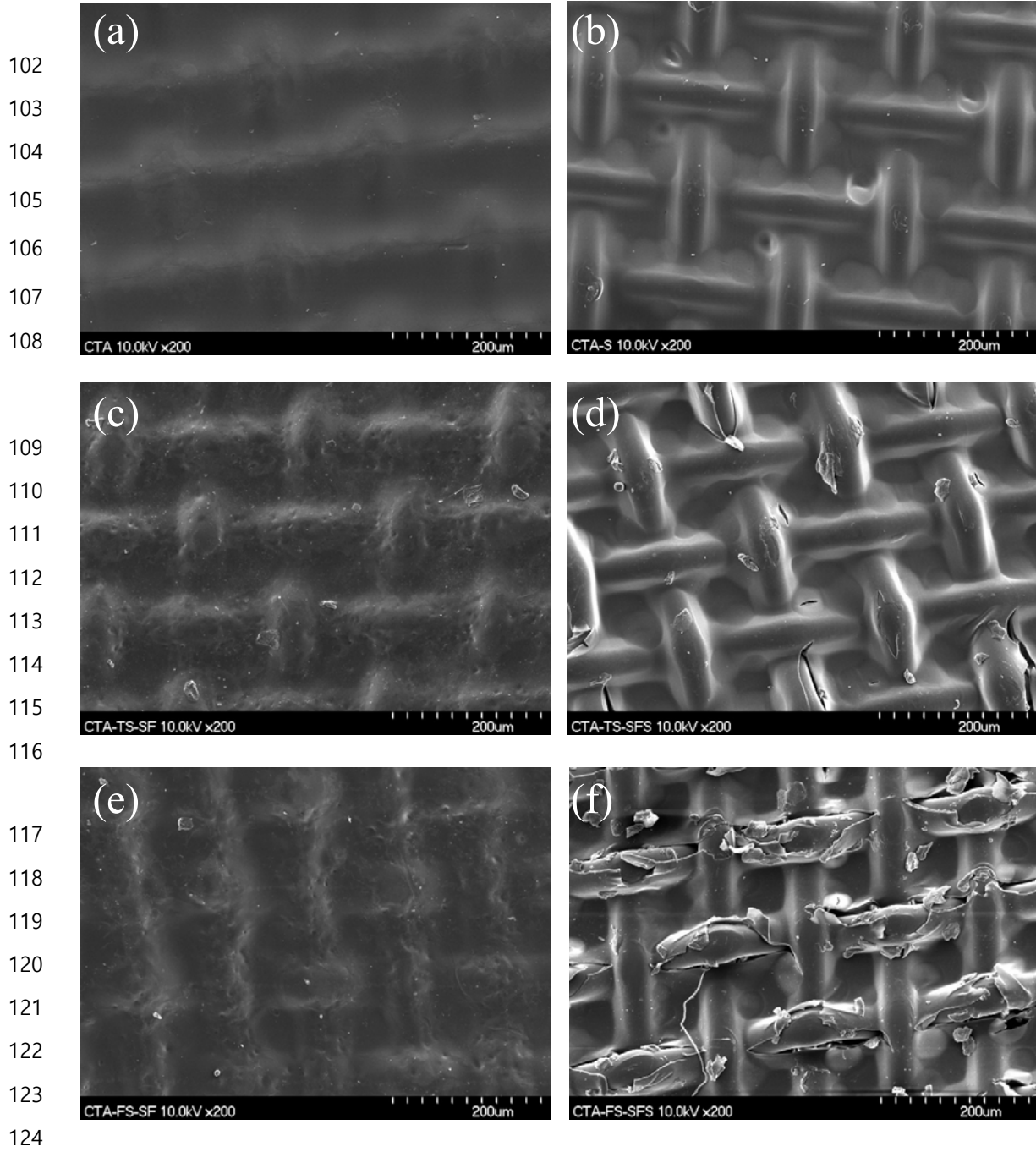


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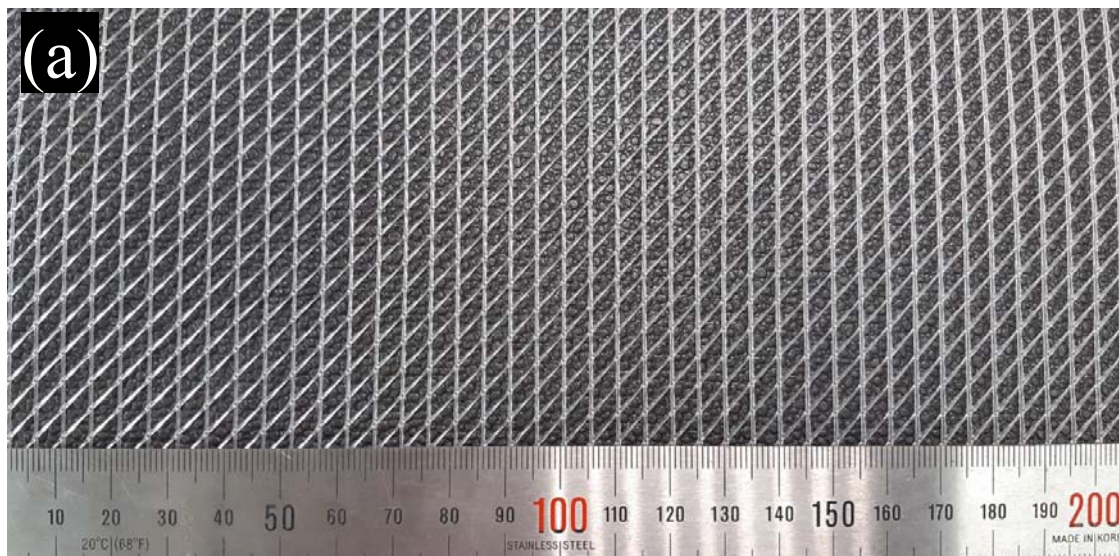
Figure S2. SEM images of active (a, c and e) and support (b, d and f) layer surfaces (PA-TFC membrane): before pressurization (a and b), after pressurization using tricot spacer (TS) (c and d) and feed spacer (FS) (e and f).



125 Figure S3. SEM images of active (a, c and e) and support (b, d and f) layer surface (CTA-ES2
126 membrane): before pressurization (a and b), after pressurization using tricot spacer (TS) (c and
127 d) and feed spacer (FS) (e and f).

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149 Figure S4. Images of feed spacer (a) and tricot spacer (b)

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158 Table S1. Mechanical properties of PA-TFC and CTA-ES2 membrane in dried and wet state

Properties	PA-TFC		CTA-ES2	
	Dry state	Wet state	Dry state	Wet state
Ultimate tensile strength (UTS, MPa)	66.2±3.18	66.6±3.94	47.6±0.8	44.1±1.2
Elastic modulus (MPa)	716.7±57.26	736.3±74.87	824.8±167.47	531.8±102.25
Elongation at break (%)	15.4±1	14.6±0.275	12.1±0.78	11.4±1.33

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Experimental setup

162 The cross-flow FO experimental setup used in this study is identical to the one described
163 in our previous studies⁹⁻¹¹. An acryl membrane cell is consisted of two flow channels
164 separated by membrane with dimensions of 2.5 cm (W) x 7.5 cm (L) x 0.3 cm (H) that
165 yield effective membrane area of 18.75 cm². Pressure gauges with precision of ± 0.0005
166 bar precision were installed at the inlet and outlet of the feed and draw channel in order to
167 measure and monitor pressure drop across the feed and draw channels. The effective
168 applied pressure (ΔP) was postulated as the average between inlet and outlet pressure of
169 each channel. The draw solutions were made using NaCl (99.7 % purity, OCI, Korea) as
170 concentrations of 0.0476, 0.0952, 0.1428, 0.1904, 0.238, 0.2856 and 0.3332 M which are
171 equivalent to osmotic pressures of 2 ~ 14 bars at 2 bar interval. A conductivity meter was
172 set in the feed solution to monitor concentration change by the water presence and the
173 reverse solute flux. Water volume of all the solutions are fixed as 2 L and an electronic
174 mass balance (GF-6100, A&D Company, Japan) was used to record the variation in the

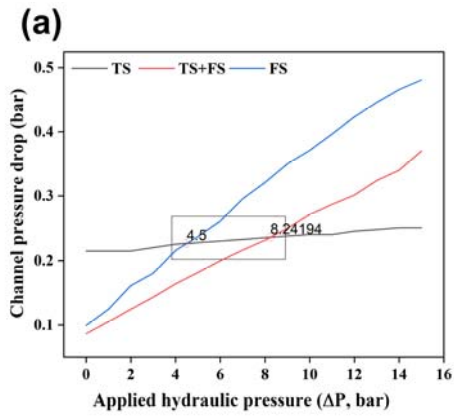
175 water mass to enable the calculation of the water flux. Each solution was circulated with
176 constant flowrate of 300 ml/min (i.e. 6.66 cm/s in cross flow velocity) using the pump
177 drive (EW-75211-10, Cole-Parmer, USA) fitted with a magnetic pump head (GB-
178 P25.JVS.A.B1, Micropump, USA). To pressurize the feed solution, a magnetic pump drive
179 (EW-75211-10, Cole-Parmer, USA) coupled with a magnetic pump head (GAF-T23-
180 DEMSE, Micropump, USA) was employed to generate hydraulic pressures ranging from
181 2 to 14 bar at 2 bar interval.

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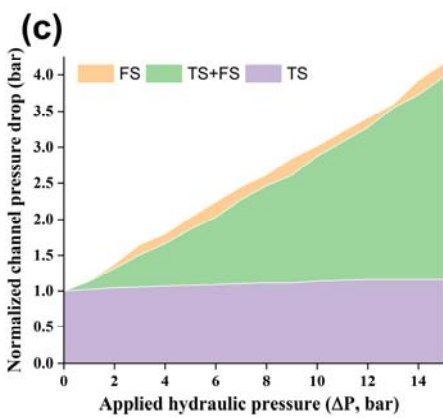
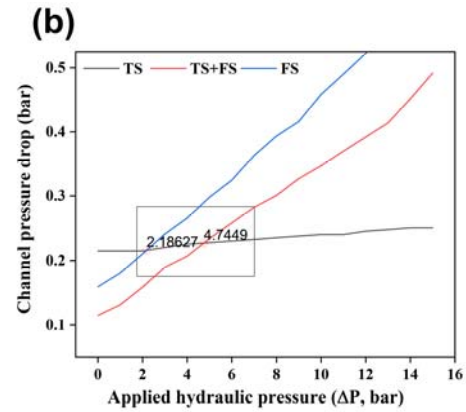
183 Channel pressure drop

184 Channel pressure drop was obtained by using difference between inlet and outlet pressure of
185 draw flow channel filled with different spacer configurations. Since it is not viable to visually
186 observe actual membrane deformation under hydraulic pressure (Even if visual observation is
187 possible reduced channel volume and geometry cannot be quantified experimentally) during
188 operation this channel pressure drop can provide useful insights as it can be indirect indicator
189 of membrane deformation. Figure S3 (a) and (b) illustrates channel pressure drop depending
190 on spacer configuration as a function of applied hydraulic pressure. In both membranes, TS+FS
191 and FS reveal steep increase compared to gradual increase of TS due to membrane stretching.
192 In TS configuration, structural characteristics of membrane does not seem affect pressure drop
193 as both membranes show almost identical pressure drop. However, extent of pressure drops by
194 membrane stretching turned out to be membrane specific. To examine this difference between
195 two membranes, Hydraulic pressure point was marked where pressure drop of TS+FS and FS
196 exceed that of TS. In PA-TFC, 4.5 and 8.241 bar of pressure drop were required for TS+FS and
197 FS respectively to match identical level of pressure drop of TS while CTA-ES2 membrane
198 demand 2.186 and 4.744 that is approximately half of hydraulic pressure of PA-TFC. To
199 identify ratio of pressure, drop change, in Figure S3 (c) and (d), pressure drop was normalized
200 with the pressure drop under non-pressurized condition. While TS shows almost identical ratio
201 of pressure drop change, steep increase of normalized pressure was identified in both
202 membranes. PA-TFC reveals slightly lower ratio compared to CTA-ES2 membrane with
203 smaller deviation between TS+FS and FS.

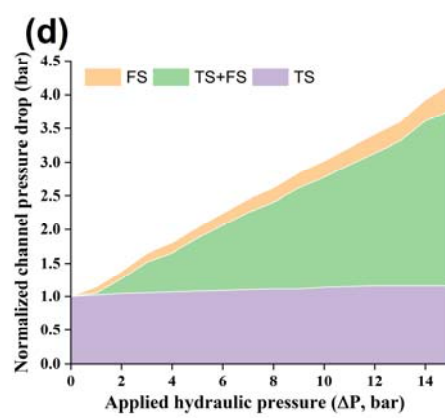
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207 Figure S5. Channel pressure drop and normalized channel pressure drop of flow channels filled
 208 with different spacer configurations for (a), (c) PA-TFC and (b), (d) CTA-ES2

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210 Theoretical backgrounds

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212 Characterization of FO membrane associated with effect of spacer

213 Since early 2000's there have been numerous research efforts to set universal criteria for FO
214 membrane performance along with elucidating permeation mechanism of ODMPs. Since there
215 is no commercialized membrane specifically made for other ODMPs (i.e. PRO and PAFO) at
216 the current stage this FO membrane and module are commonly used for all ODMPs. For
217 characterization of FO membranes, three membrane performance parameters are commonly
218 used based on conventional S-D model: Water permeability, solute permeability and structural
219 parameter, so called A, B and S values¹. These apparent A, B and S values are determined by
220 various experimental methods and theoretical approaches.

221 Unlike pressure-driven processes where effect of pressure is dominant and pronounced,
222 ODMPs have more sensitive and complex permeation mechanism that creates many
223 discrepancies in measured performance parameters among FO membrane studies. These
224 discrepancies are mainly attributed to two factors: (1) concentration polarization (CP) (2)
225 membrane deformation. While Internal Concentration Polarization (ICP) is normally
226 considered to be not affected by hydrodynamic condition several experimental studies showed
227 extent of ICP can be altered with External Concentration Polarization (ECP) by spacer
228 configuration and cross-flow velocity^{2,3}. To take into account ICP and ECP effect, many
229 theoretical models^{4,5} have been suggested based on mass balance across the membrane layers
230 and mass transfer coefficient of flow channel (i.e. one dimensional lump parameter derived
231 from empirical correlation between dimensionless numbers)⁶.

232 The other factor causing discrepancy among literatures is that conventional characterization

233 method utilizes pressurized condition while ODMPs are operated either under non-pressurized
234 condition or pressure does not act as a main driving force in processes. For this reason, non-
235 pressurized methods for FO membrane was suggested, postulating that ICP can be minimized
236 when active layer of membrane faces draw side^{7,8}. These new methods seem to be relatively
237 well accepted in FO community by virtue of better rationale, however, these methods have
238 inherent limitation as it is not able to resolve effect of spacer-induced membrane deformation.

239 In efforts to further elucidate characteristics of FO membrane in association with structural
240 deformation, novel characterization method for FO membrane was introduced based on
241 irreversible thermodynamics and reciprocal relations of hydraulic and osmotic pressures⁹.
242 Since irreversible thermodynamic model regards membrane as a black box mass transport
243 through membrane is determined by four phenomenological coefficients without consideration
244 of membrane's structural characteristics. However, this approach can provide interesting
245 insight in regard to membrane's structural vulnerability towards external stress created by
246 hydraulic pressure and supporting spacer. More particularly, deterioration of membrane's
247 selectivity and resultant change of interdependence between water and solute transports can be
248 quantified as a function of hydraulic pressure.

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260 Solution-diffusion model

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262 General flux equations for water flux (J_w) and solute flux (J_s) based on conventional S-D model
263 can be expressed as below:

$$264 \quad J_w = A(\Delta P - \Delta\pi) \quad (S1)$$

$$265 \quad J_s = B \cdot \Delta\pi \quad (S2)$$

266 Where A and B are water and solute permeability coefficient, ΔP and $\Delta\pi$ are hydraulic and
267 osmotic pressure differences respectively.

268 Here, ΔP is approximately zero in FO, negative in PRO and positive in PAFO. The driving
269 force for solute flux through the selective layer is concentration difference across it,
270 proportional to $\Delta\pi$.

271 Based on general flux equations, below equations can be derived with account of internal
272 concentration polarization (ICP) and external concentration polarization (ECP) in associated
273 with membrane structural properties and hydrodynamics near membrane surfaces⁷:

$$274 \quad J_w = A \left[\frac{\pi_D \exp\left(-\frac{J_w S}{D}\right) - \pi_F \exp\left(-\frac{J_w}{k}\right)}{1 + \frac{B}{J_w} \left[\exp\left(\frac{J_w}{k}\right) - \exp\left(\frac{J_w S}{D}\right) \right]} \right] \quad (S3)$$

$$275 \quad J_s = B \left[\frac{C_D \exp\left(-\frac{J_w S}{D}\right) - C_F \exp\left(\frac{J_w}{k}\right)}{1 + \frac{B}{J_w} \left[\exp\left(\frac{J_w}{k}\right) - \exp\left(-\frac{J_w S}{D}\right) \right]} \right] \quad (S4)$$

276 Where C_D , C_F are bulk concentrations and π_D and π_F are corresponding osmotic pressure
277 on the draw and feed side respectively. k is the solute mass transfer coefficient and D is the

278 bulk diffusion coefficient of the draw salt.

279 In eqs (3), (4), $\exp\left(-\frac{J_w S}{D}\right)$ resolves ICP in association with structural parameter and

280 $\exp\left(\frac{J_w}{k}\right)$ resolves ECP in association with mass transfer coefficient.

281 S is the structural parameter of the support layer to diffusion within the porous support,

282 determined by

$$283 \quad S = \frac{t_s \tau}{\varepsilon} \quad (S5)$$

284 where τ , t and ε are the tortuosity, thickness and porosity of the support layer.

285 mass transfer coefficient, k, is determined by eqs. (S6) and (S7). Eqs (S7) is valid only in

286 laminar flow regime, which is applied for most membrane channel studies.

$$287 \quad k = \frac{ShD}{d_h} \quad (S6)$$

$$288 \quad Sh = 1.85 \left(Re Sc \frac{d_h}{L} \right)^{0.33} \quad (S7)$$

289 Where Sh, Re, Sc are sherwood number, reynolds number and schmidts number. d_h and L are

290 hydraulic diameter and channel length. Eq. (7) is valid only in laminar flow regime, which is

291 used in most membrane channel studies.

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296 Irreversible thermodynamic model

297 Irreversible thermodynamic models^{12,13} are rooted in four phenomenological coefficients and
298 Onsager reciprocity relations¹⁴. General irreversible thermodynamic model can be expressed
299 as below:

$$300 \quad J_v = L_p \Delta p + L_{pD} \Delta \pi \quad (\text{S8})$$

$$301 \quad J_D = L_{Dp} \Delta p + L_D \Delta \pi \quad (\text{S9})$$

$$302 \quad L_{Dp} = L_{pD} \quad (\text{S10})$$

303 where, J_v is the volumetric flux, J_D is the relative solute velocity to solvent, L_p is the
304 volumetric transport coefficient, L_{Dp} and L_{pD} are the equivalent reciprocal coefficients from
305 Onsager, and L_D is the diffusive solute transport coefficient.

306 When net flux through membrane is zero ($J_v = 0$) staverman reflection coefficient can be
307 described by eq. (11). Using eq. (1) with null-flux condition eq. (S12) can be developed to
308 obtain reciprocal coefficient.

$$309 \quad \sigma = \left(\frac{\Delta P}{\Delta \pi} \right)_{J_v=0} \quad (0 \leq \sigma \leq 1) \quad (\text{S11})$$

$$310 \quad \Delta P = - \frac{L_{pD}}{L_p} \Delta \pi \quad (\text{S12})$$

311 It should be noted that J_D is not solute flux (\dot{n}_s), which is more generally used for solute
312 transport. Solute flux (\dot{n}_s) can be defined in terms of J_v , J_D and c_s in below eqs (S13)

$$313 \quad \dot{n}_s = (J_v + J_D)c_s \quad (J_v = 0) \quad (\text{S13})$$

314 Where c_s is concentration of solute in NaCl solution.

315 Under null-flux mode, J_D is naturally determined by normalizing solute flux with respective

316 NaCl concentration.

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